

1D and 2D experiments set up by NMR

by 林永峰 (Y.F.)

GRC NMR Core Facility

Experiments set up in GRC

- ☀ **1D ^1H NMR**
- ☀ **1D ^{13}C NMR**
- ☀ **2D homonuclear NMR**
- ☀ **2D heteronuclear NMR**
- ☀ **Measurement of coupling constants**
- ☀ **DOSY**

1. 1D ^1H NMR

☀ ^1H NMR (一般氫光譜實驗)

IGRC_ID_1H

IGRC_ID_1H-homodecouple

☀ Solvent Suppression (溶劑峰壓制實驗)

IGRC_ID_1H-zgpr

IGRC_ID_1H-zgcppr

Multiple Solvent Suppression

☀ Selective Excitation (選擇性激發實驗)

sel_COSY

sel_TOCOSY

sel_NOESY

sel_ROESY

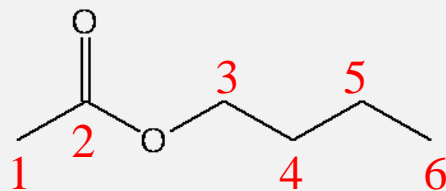


Button NMR set up

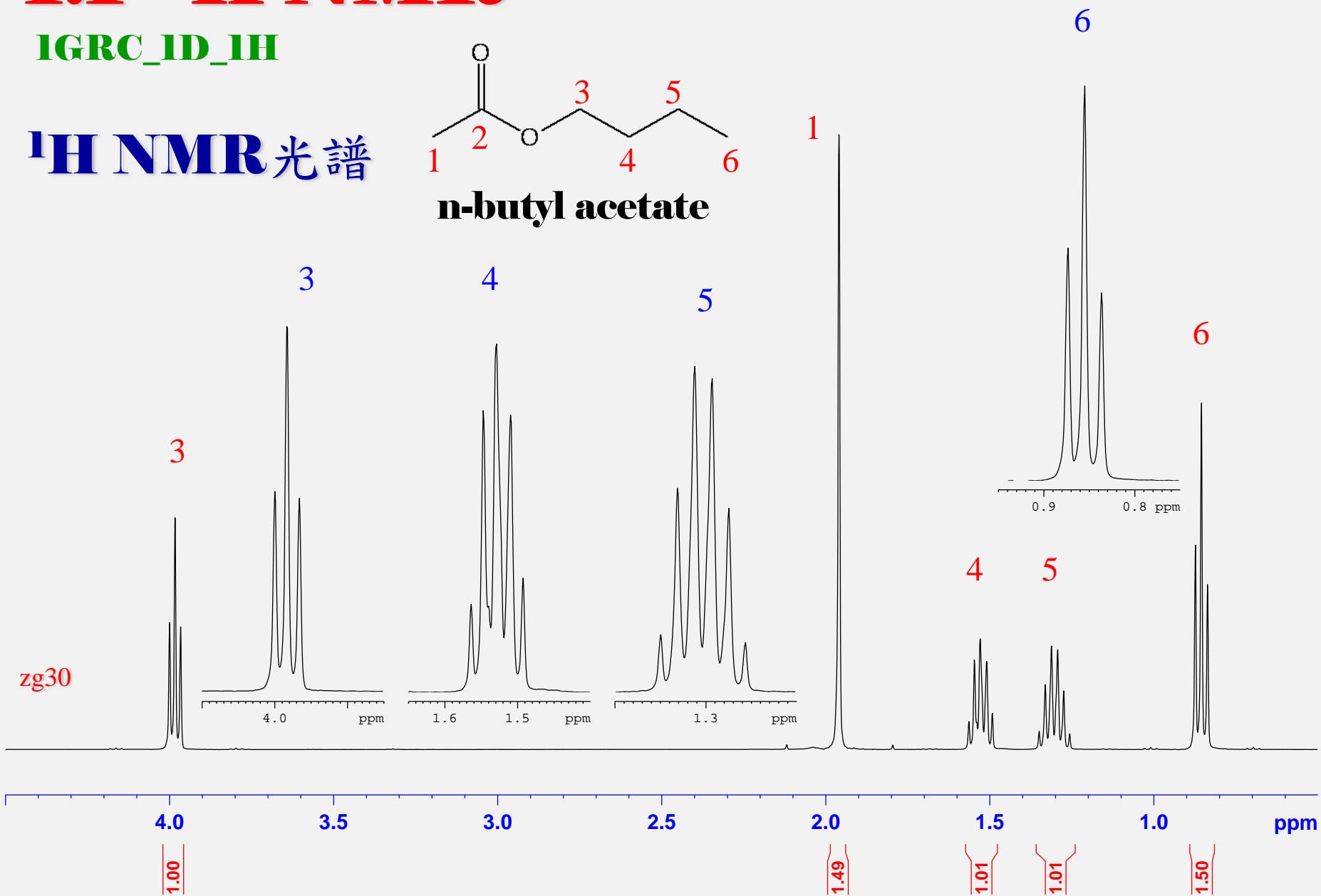
1.1 ^1H NMR

IGRC_ID_1H

^1H NMR 光譜

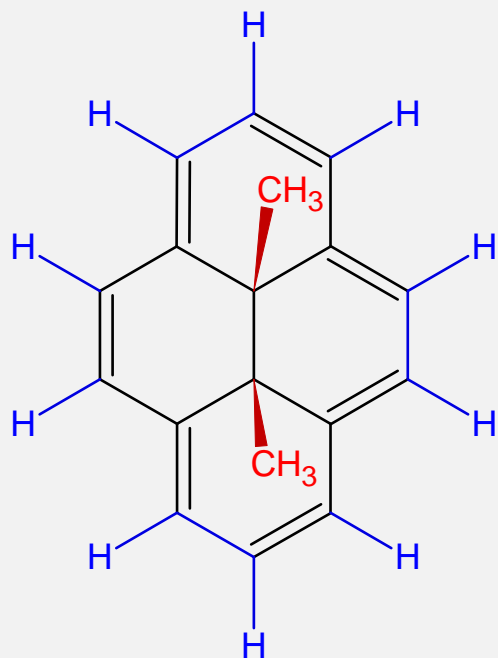


n-butyl acetate



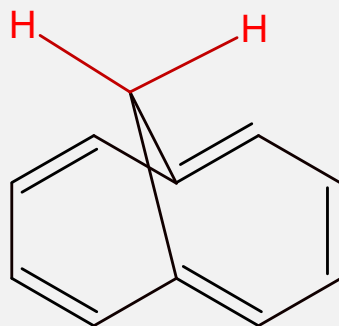
1.1.1 ^1H NMR

☀ Anisotropy effect



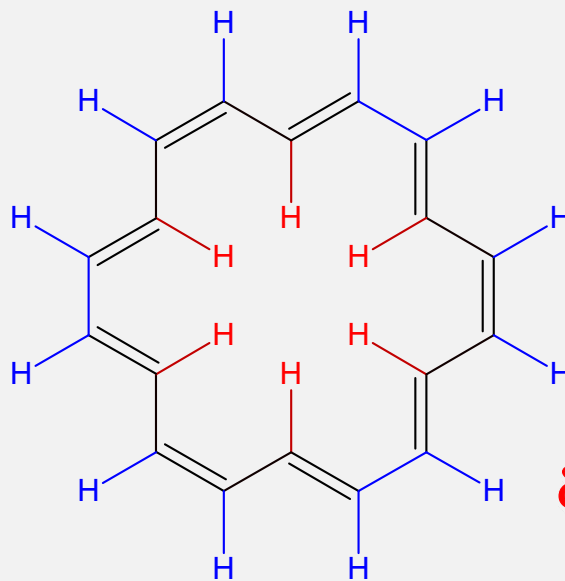
$\delta_{\text{ring}} \rightarrow 8.14-8.64$ ppm

$\delta_{\text{Me}} \rightarrow -4.25$ ppm



$\delta_{\text{ring}} \rightarrow 7.27-6.95$ ppm

$\delta_{\text{Me}} \rightarrow -0.51$ ppm



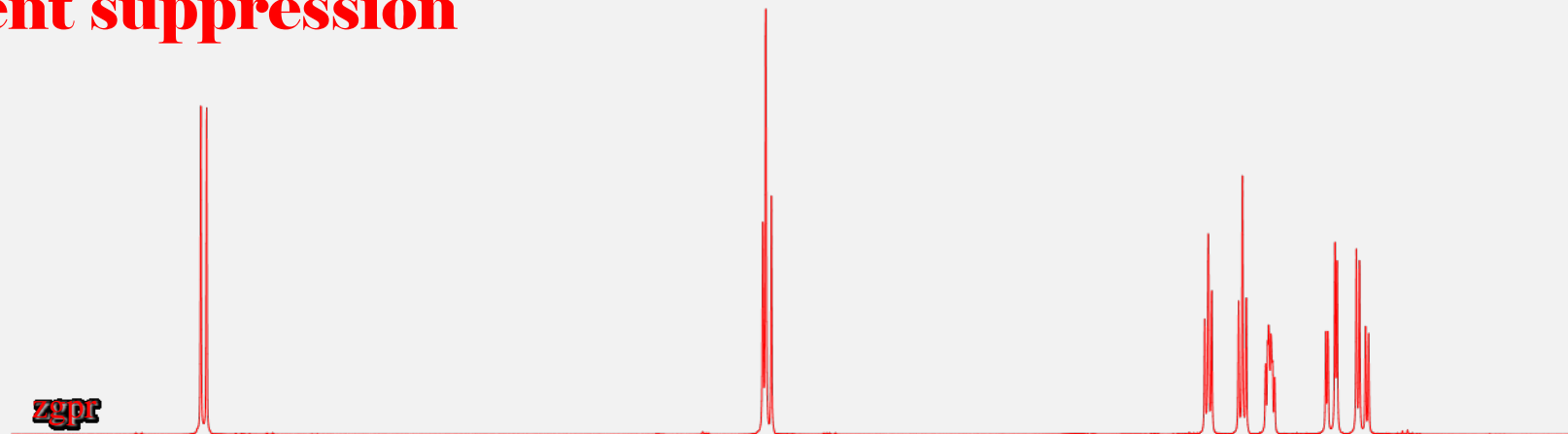
$\delta_{\text{OUTSIDE}} \rightarrow 9.28$ ppm

$\delta_{\text{INSIDE}} \rightarrow -2.99$ ppm

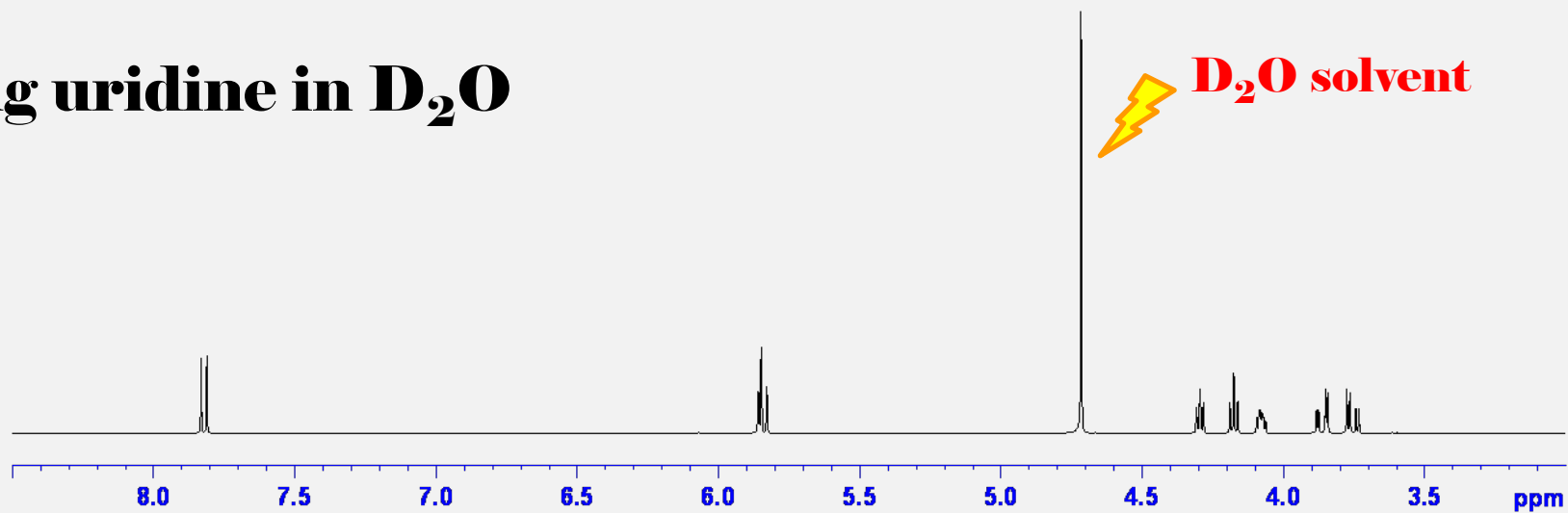
1.2 Solvent Suppression

IGRC_ID_1H-zgpr

solvent suppression



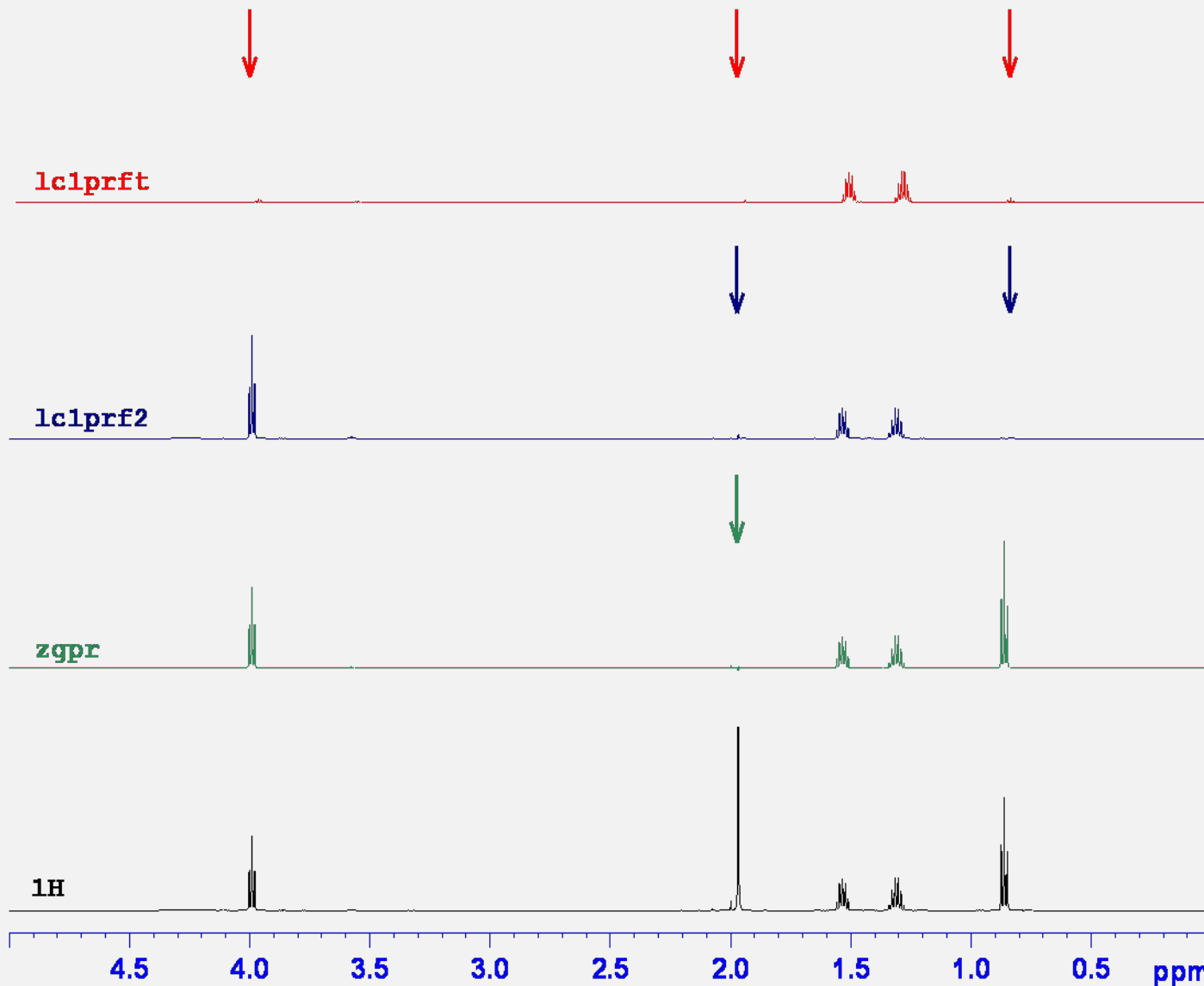
25mg uridine in D₂O



1.3 Multiple Solvent Suppression

- ☀ 利用不同的頻道進行溶劑訊號之壓制(一般實驗)
 - lclprf2(2), lclprft(3), lclpnf2(2), lclpnft(3), lclpncwfd(2) (括號內為壓制溶劑訊號之數目)
- ☀ 利用 **shape pulse** 進行溶劑訊號之壓制(PS系列)
 - zgps, lclpnps, lclpncwps
- ☀ 利用 **WET scheme** 進行溶劑訊號之壓制(WET系列)
 - wet, wetdc, wetdw

Multiple Solvent Suppression

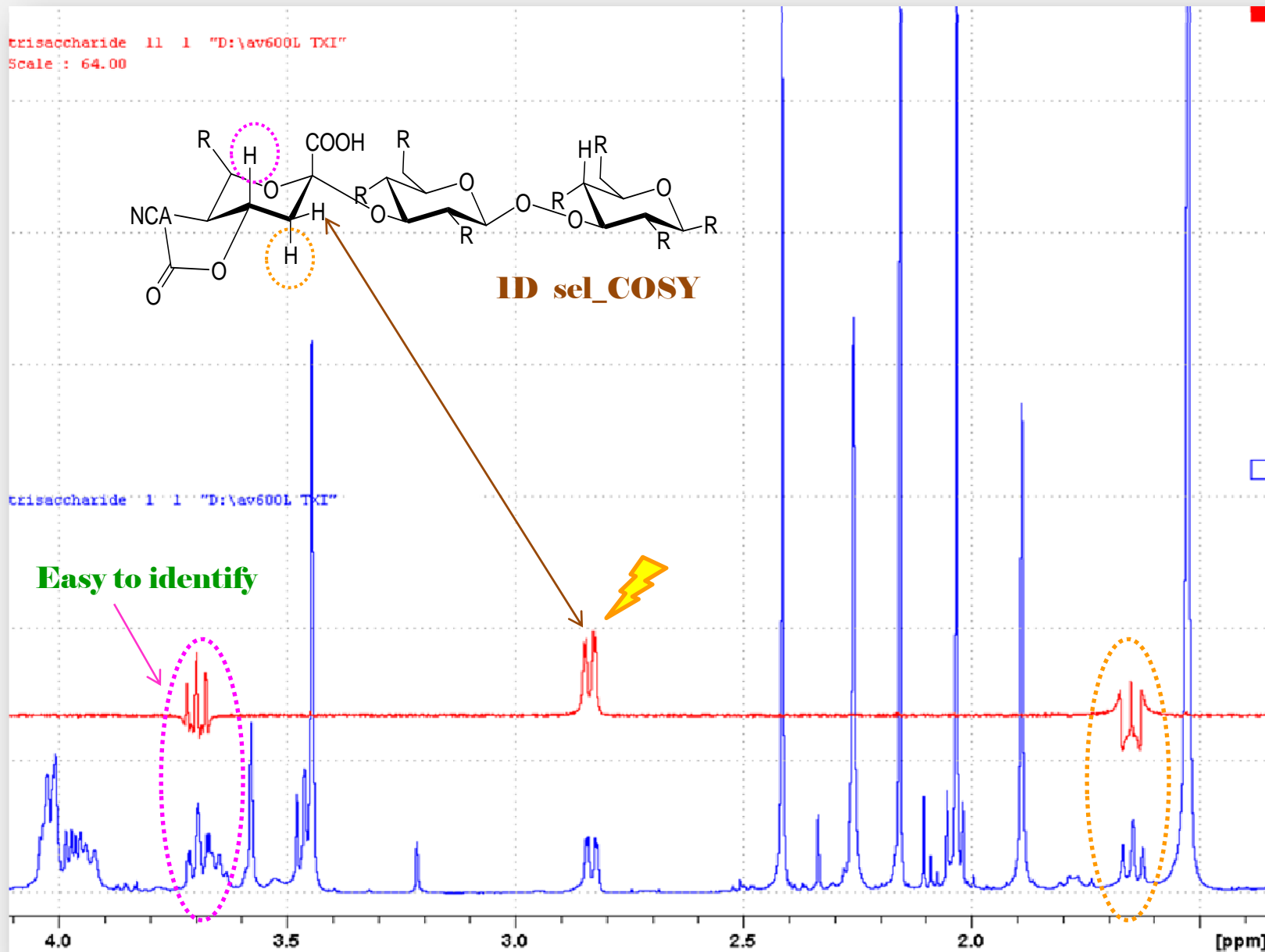


1.5 1D Selective Excitation

- ☀ **COSY, TOCSY, NOESY, ROESY** 四種實驗
- ☀ 針對特定 **peak**，觀察其相關性
- ☀ 減少實驗維度、將目標凸顯、簡化問題、減少實驗時間
- ☀ 方便低濃度樣品進行測定
- ☀ 可使用 **Button selective NMR** 軟體進行實驗

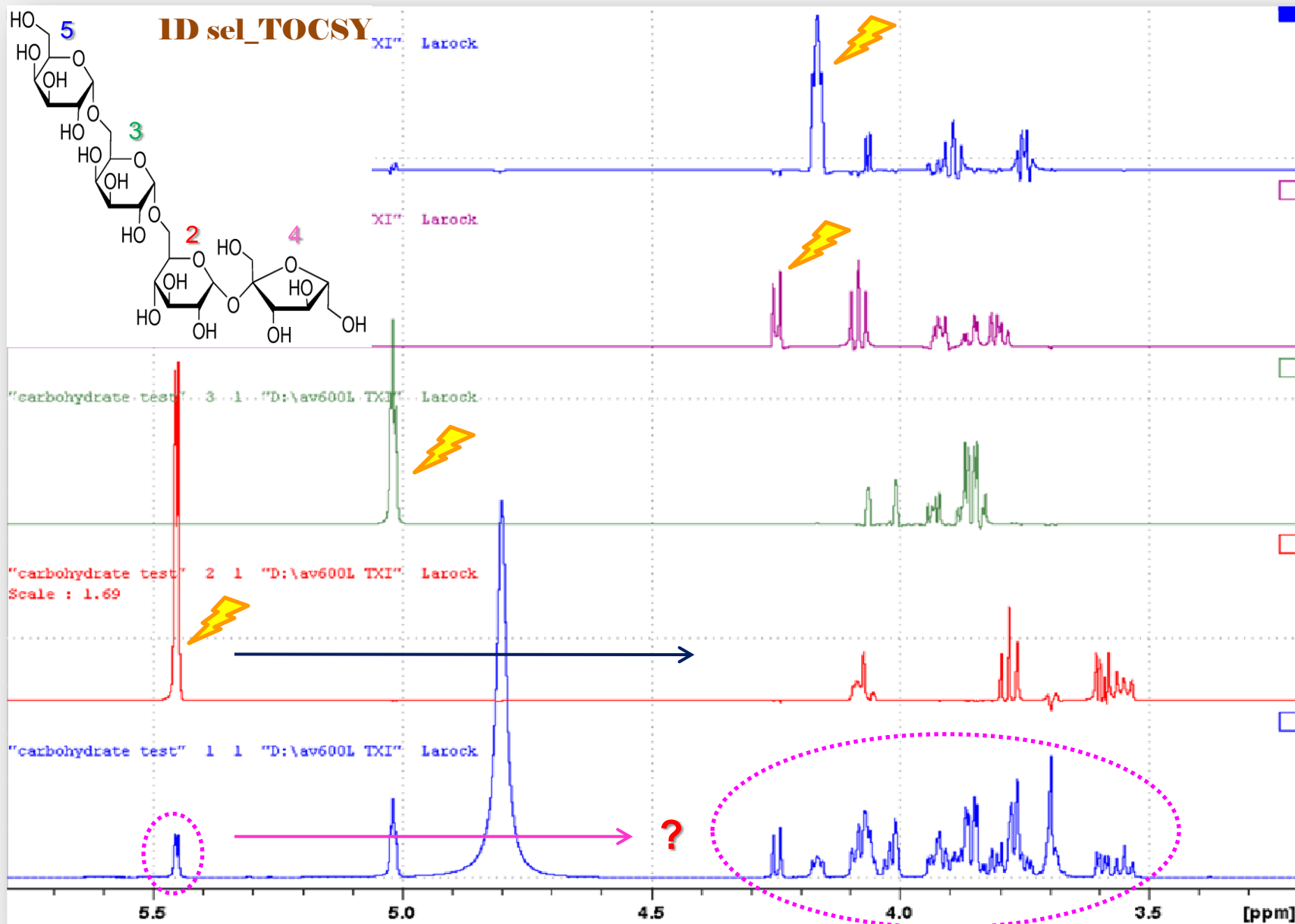
1.5.1 1D Selective COSY

Button NMR set up



1.5.2 1D Selective TOCSY

Button NMR set up



2. 1D ^{13}C NMR

☀ ^{13}C NMR (一般碳光譜實驗)

1GRC_ID_13C

1GRC_ID_13C-without decouple

1GRC_ID_13Cint

☀ Distortionless Enhancement by Polarization Transfer (區分碳的級數)

1GRC_ID_DPET45

1GRC_ID_DEPT90

1GRC_ID_DEPT135

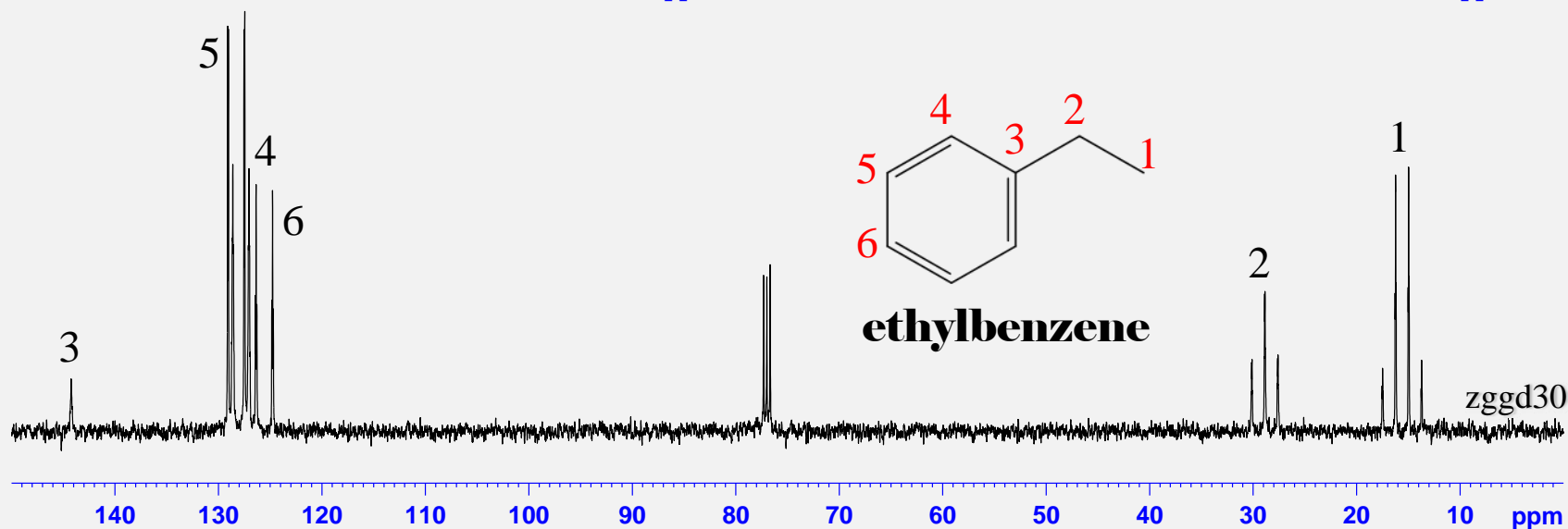
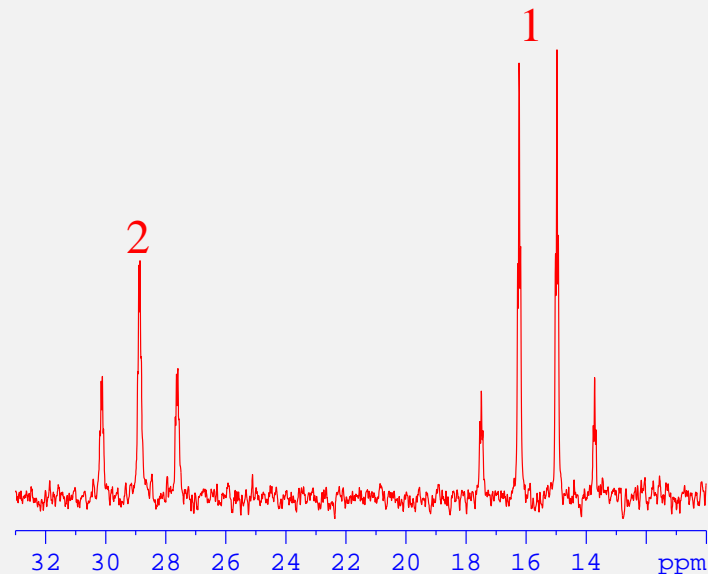
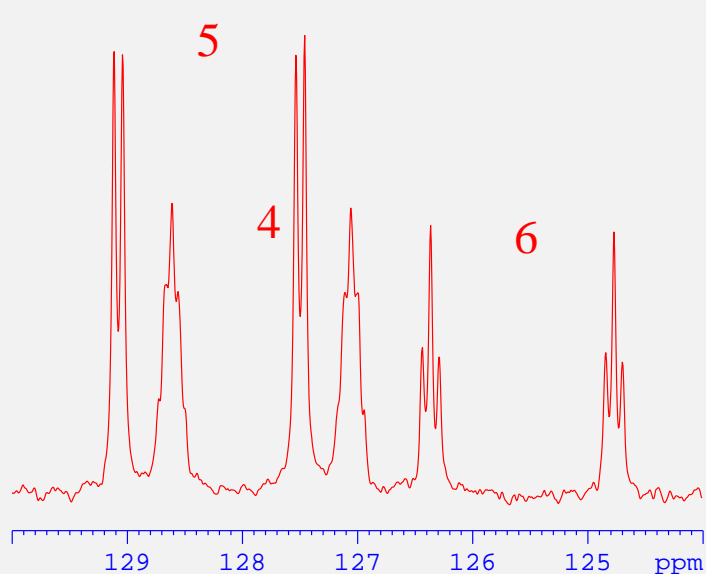
2.1 ^{13}C NMR

^{13}C 光譜特性

- ☀ 因 ^{13}C 自然含量低，則 ^{13}C - ^{13}C 彼此間偶合可忽略
- ☀ ^{13}C 會受到周圍氫原子的影響而產生分裂
- ☀ 其分裂較為複雜，而造成光譜難以判斷
- ☀ 受到直接接於碳上的氫原子之分裂較為明顯，並符合 $2nI+1$ 規則
- ☀ 訊號亦會因分裂造成強度下降，一般以去偶合(**decoupling**)方式進行實驗

2.2 自旋-自旋分裂之 ^{13}C NMR

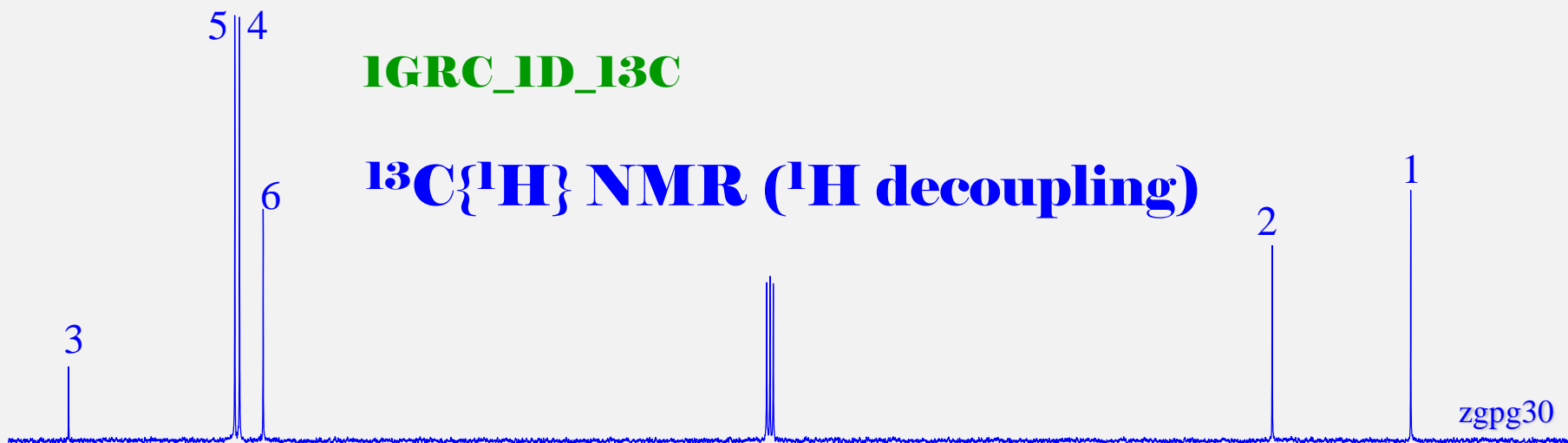
IGRC_ID_13C-without decouple



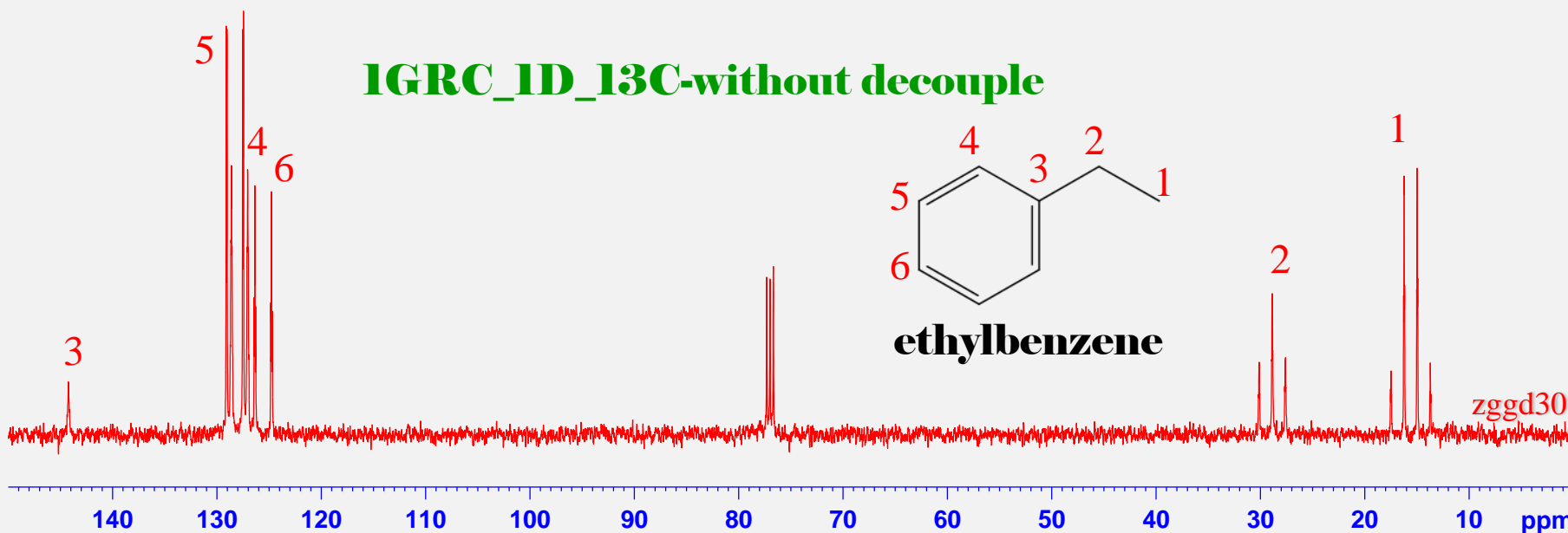
2.3 $^{13}\text{C}\{^1\text{H}\}$ NMR

IGRC_ID_13C

$^{13}\text{C}\{^1\text{H}\}$ NMR (^1H decoupling)



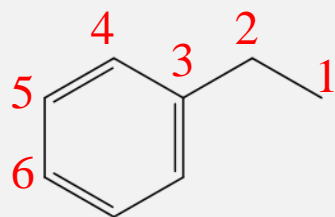
IGRC_ID_13C-without decouple



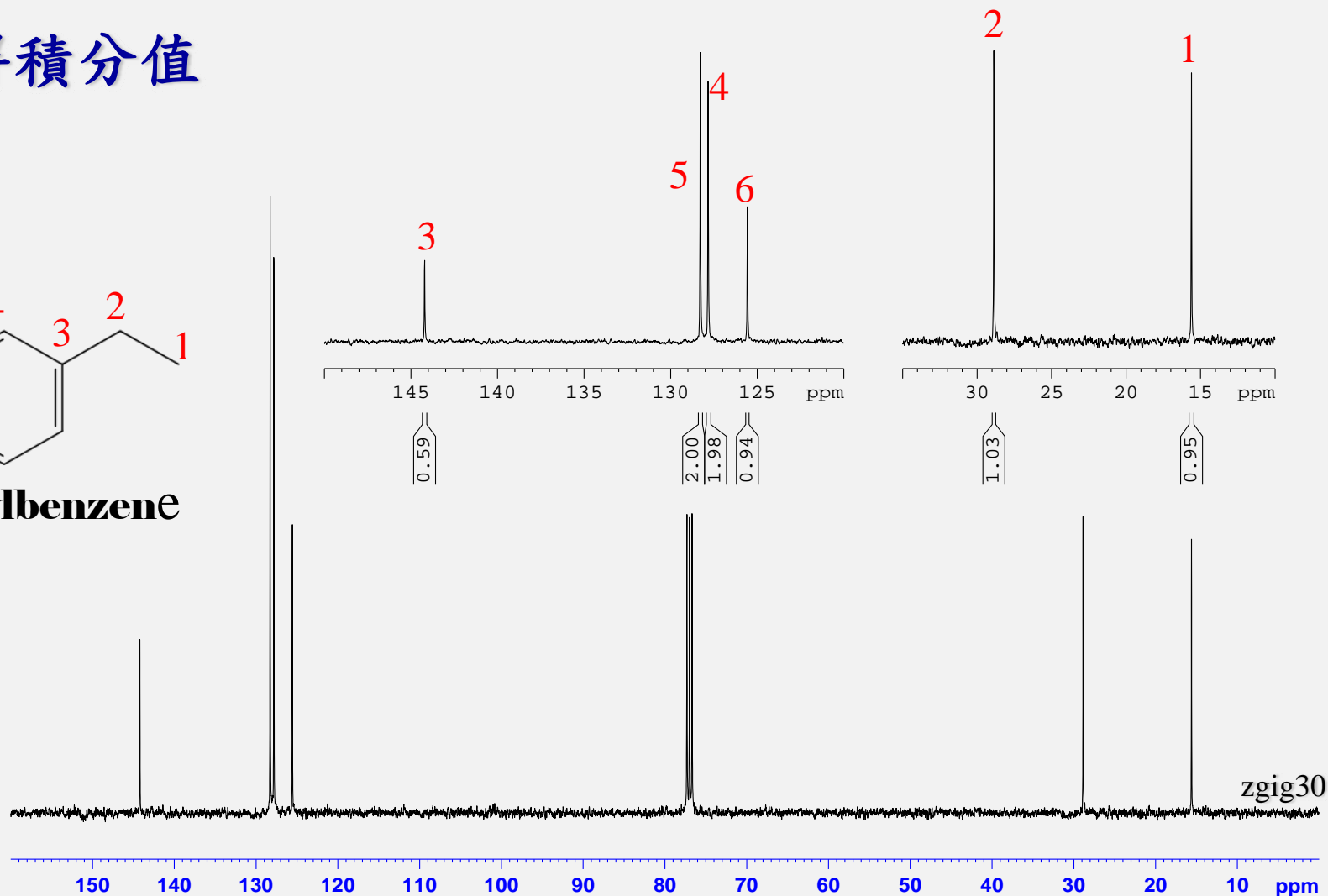
2.4 Inverse Gated Decoupling

IGRC_ID_13Cint

獲得積分値

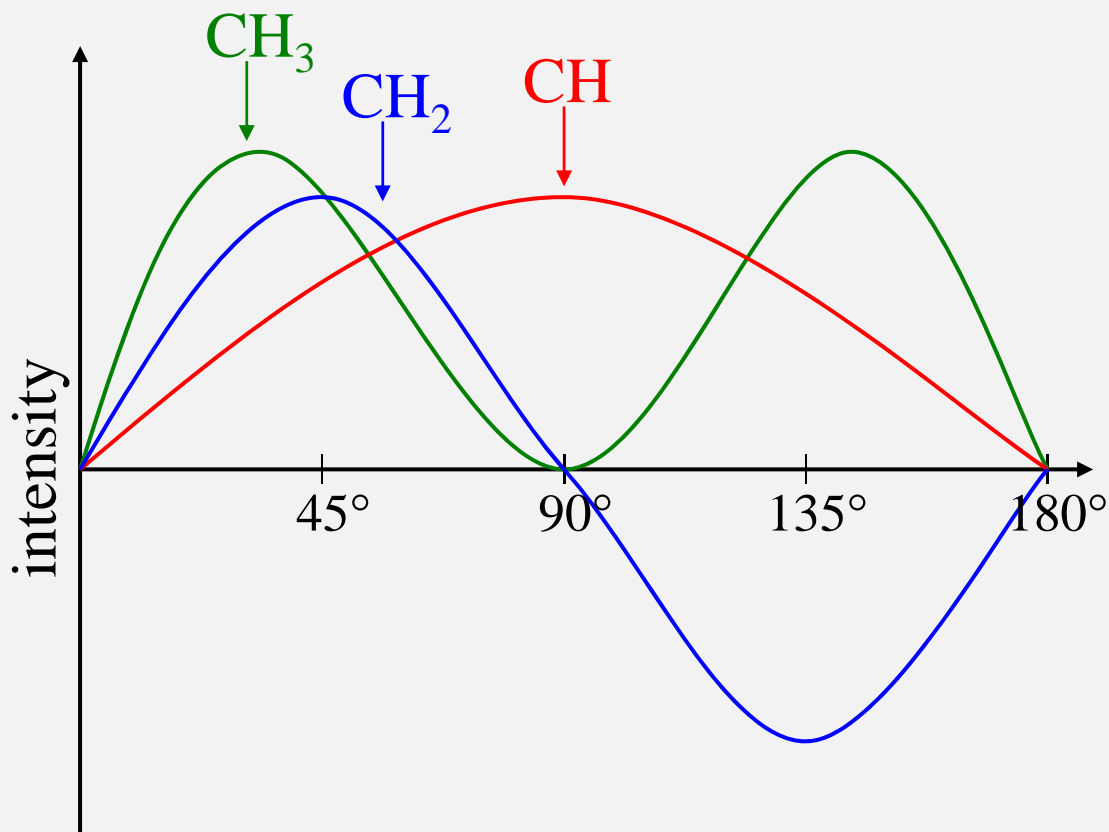


ethylbenzene



2.5.1 DEPT

Distortionless Enhancement by Polarization Transfer



IGRC_ID_DEPT45
IGRC_ID_DEPT90
IGRC_ID_DEPT135

	45°	90°	135°
CH	+	+	+
CH₂	+	×	-
CH₃	+	×	+

2.5.2 DEPT

IGRC_ID_DEPT45
IGRC_ID_DEPT90
IGRC_ID_DEPT135

區分碳的級數

DEPT135

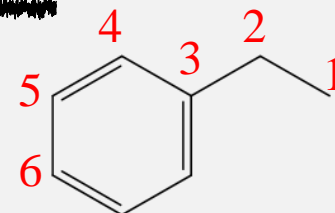
dept135

DEPT90

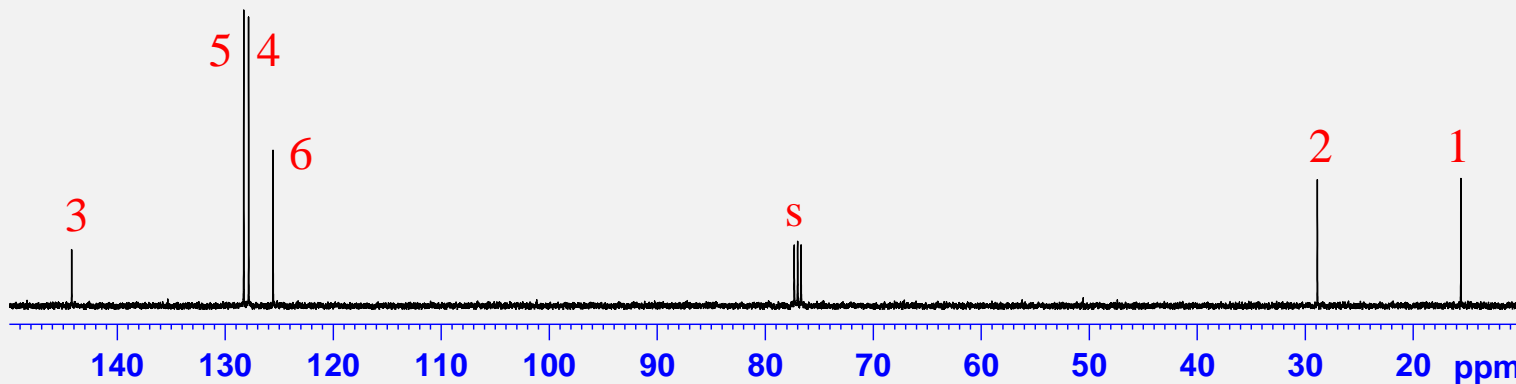
dept90

DEPT45

dept45



ethylbenzene



3. 2D homonuclear NMR

☀ COSY

IGRC_2D_COSY45
IGRC_2D_COSY90
IGRC_2D_COSY-suppression

☀ NOESY

IGRC_2D_NOESY
IGRC_2D_NOESYg
IGRC_2D_NOESY-suppression

☀ DQF-COSY

IGRC_2D_COSY-DQF
IGRC_2D_COSY-DQF-suppressin

☀ ROESY

IGRC_2D_ROESY
IGRC_2D_ROESYg
IGRC_2D_ROESY-suppression

☀ TOCSY

IGRC_2D_TOCSY_dipsi
IGRC_2D_TOCSY-mlev
IGRC_2D_TOCSY-mlevg
IGRC_2D_TOCSY-suppression

☀ INADEQUATE

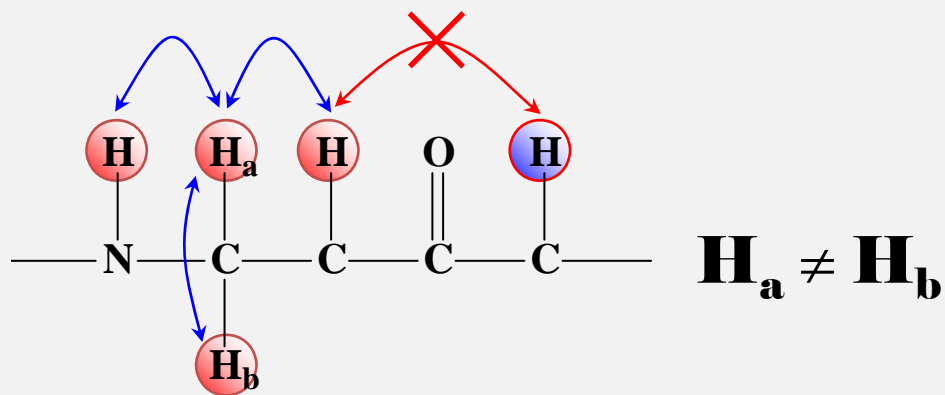
IGRC_2D_INADEQUATE

3.1 2D homonuclear COSY

☀ COSY

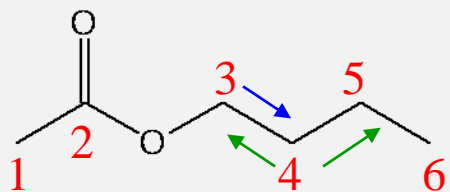
(Correlation Spectroscopy)

獲得 2J 和 3J 之關係

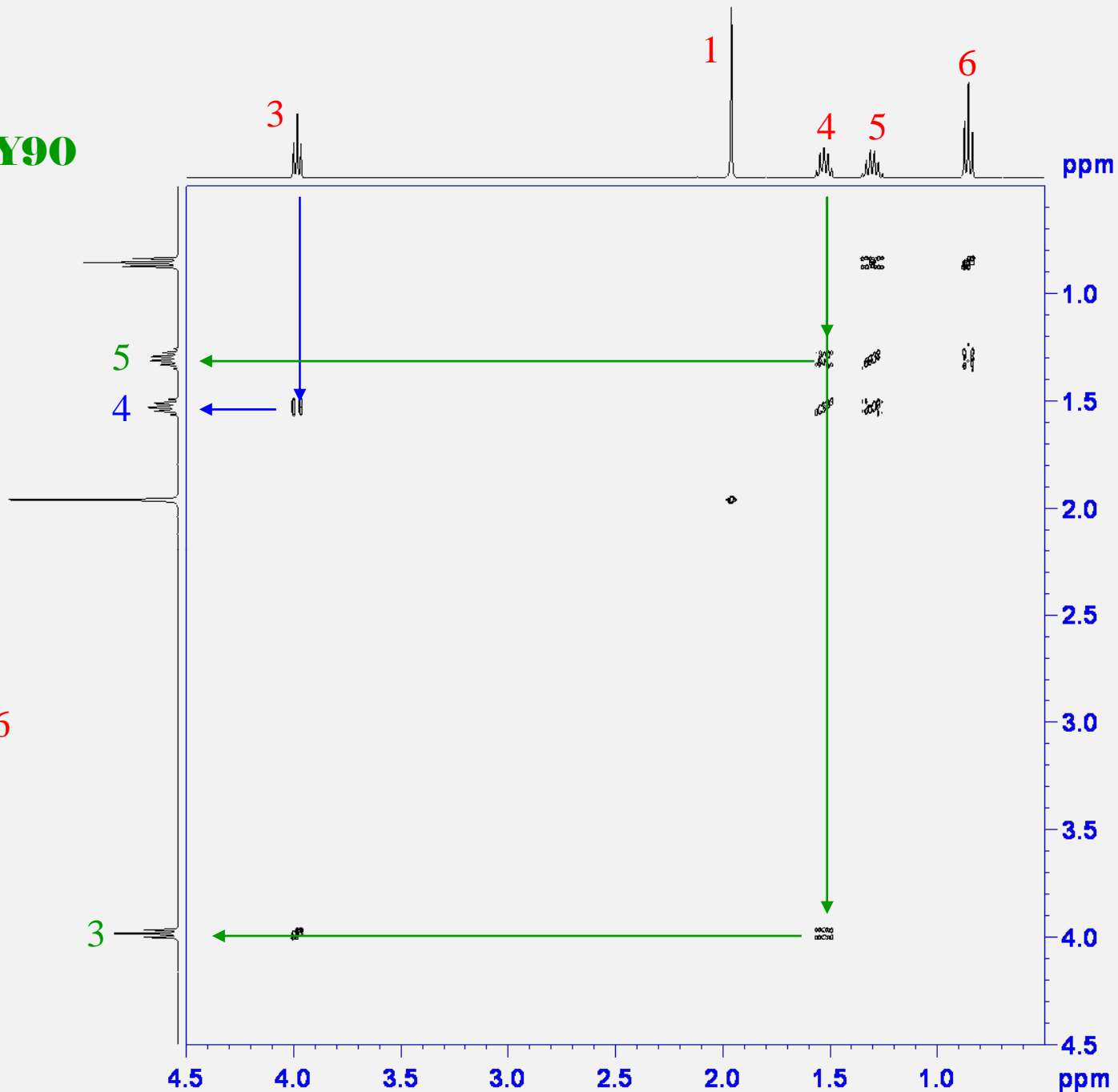


COSY

IGRC_2D_COSY90

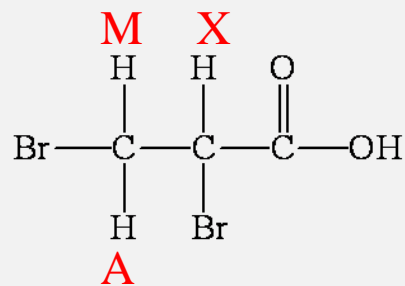


n-butyl acetate

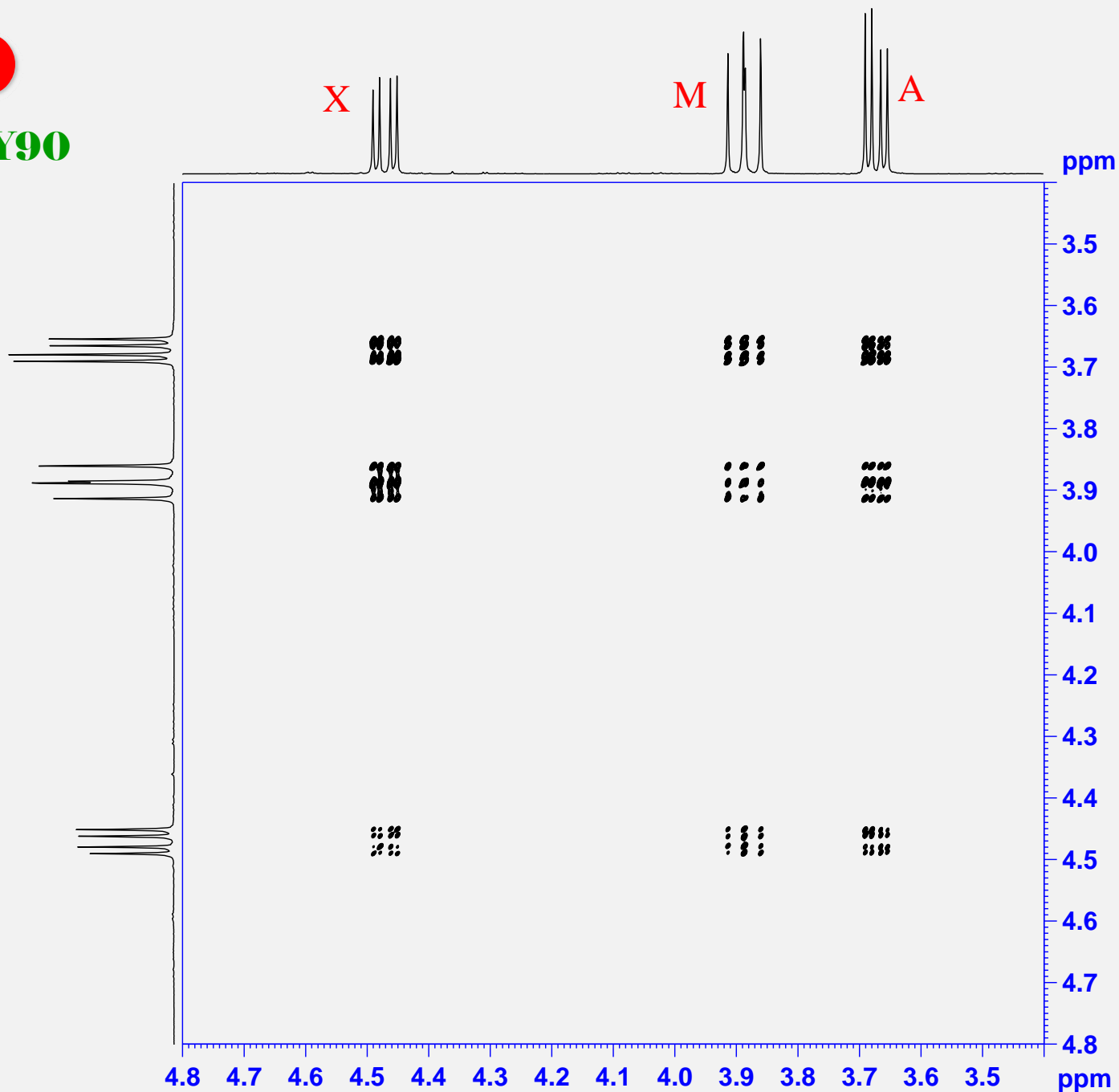


COSY90

IGRC_2D_COSY90



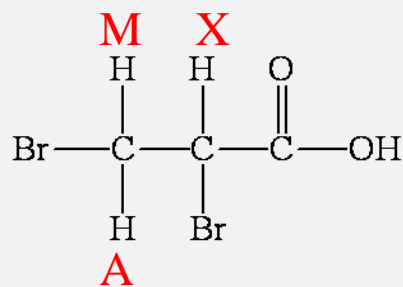
**2,3-dibromo-
propionic acid**



COSY45

IGRC_2D_COSY45

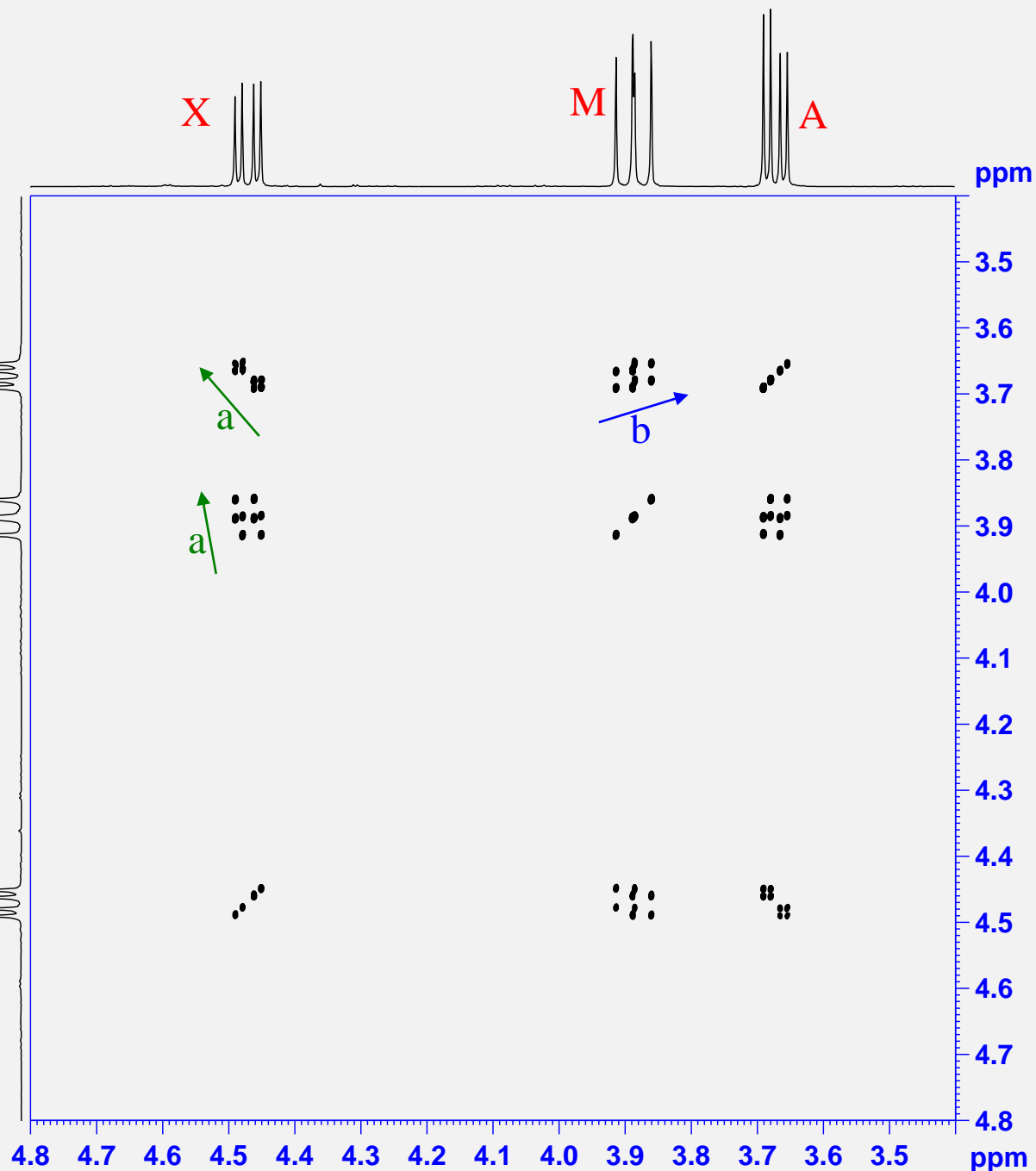
區分³J與²J耦合關係



**2,3-dibromo-
propionic acid**

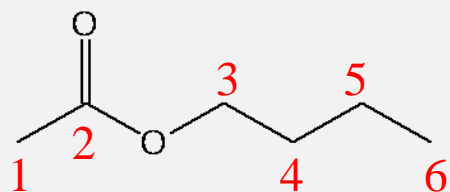
**(a) vicinal spin-spin
coupling (³J)**

**(b) geminal spin-spin
coupling (²J)**



DQF-COSY

IGRC_2D_DQF-COSY

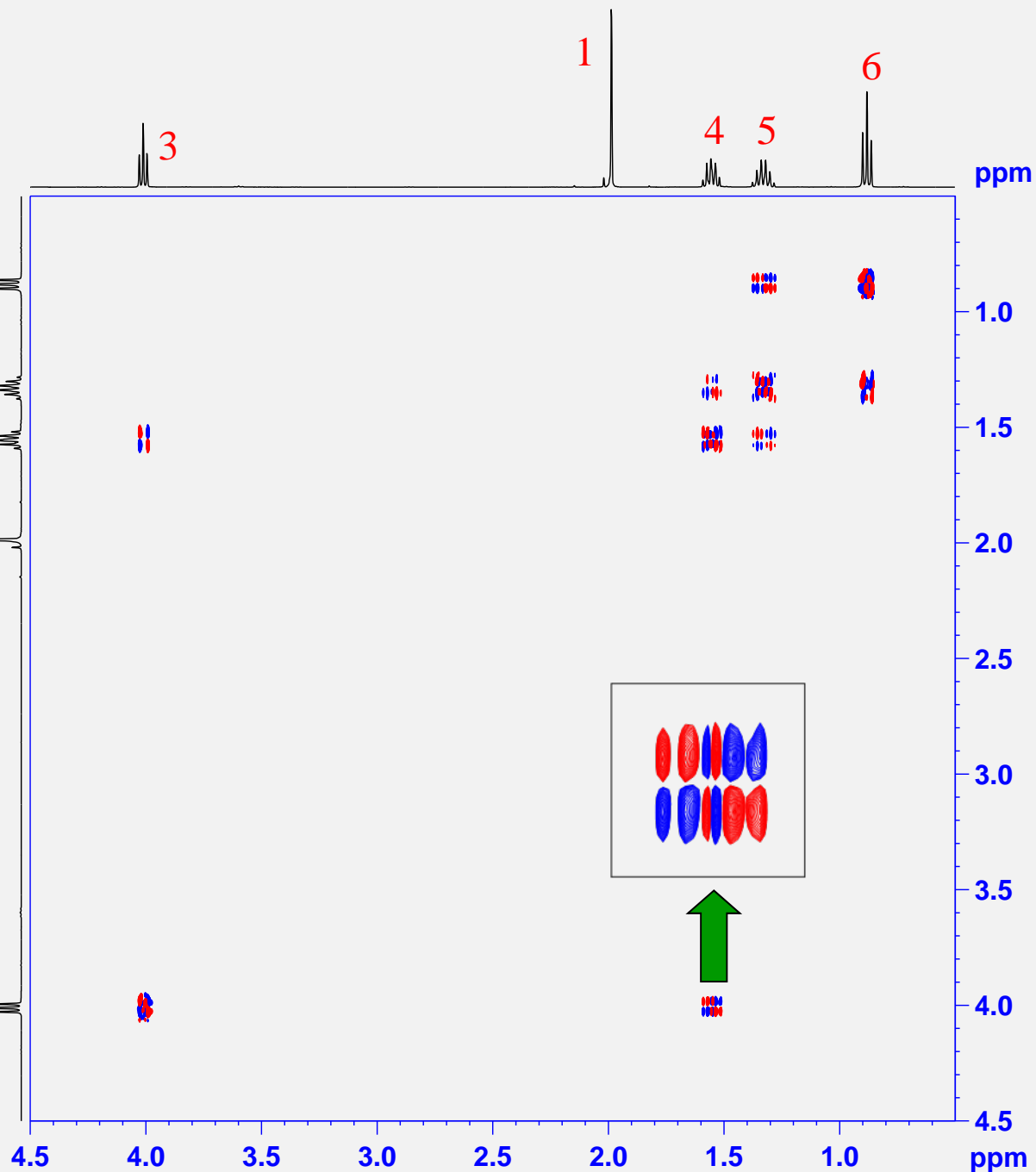


n-butyl acetate

color of the peak

positive peak

negative peak

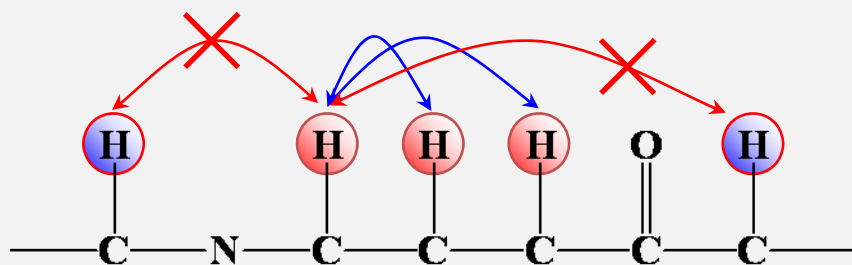


3.2 2D homonuclear TOCSY

☀ TOCSY

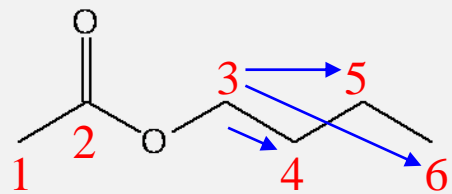
(Total Correlation Spectroscopy)

獲得所有J偶合關係

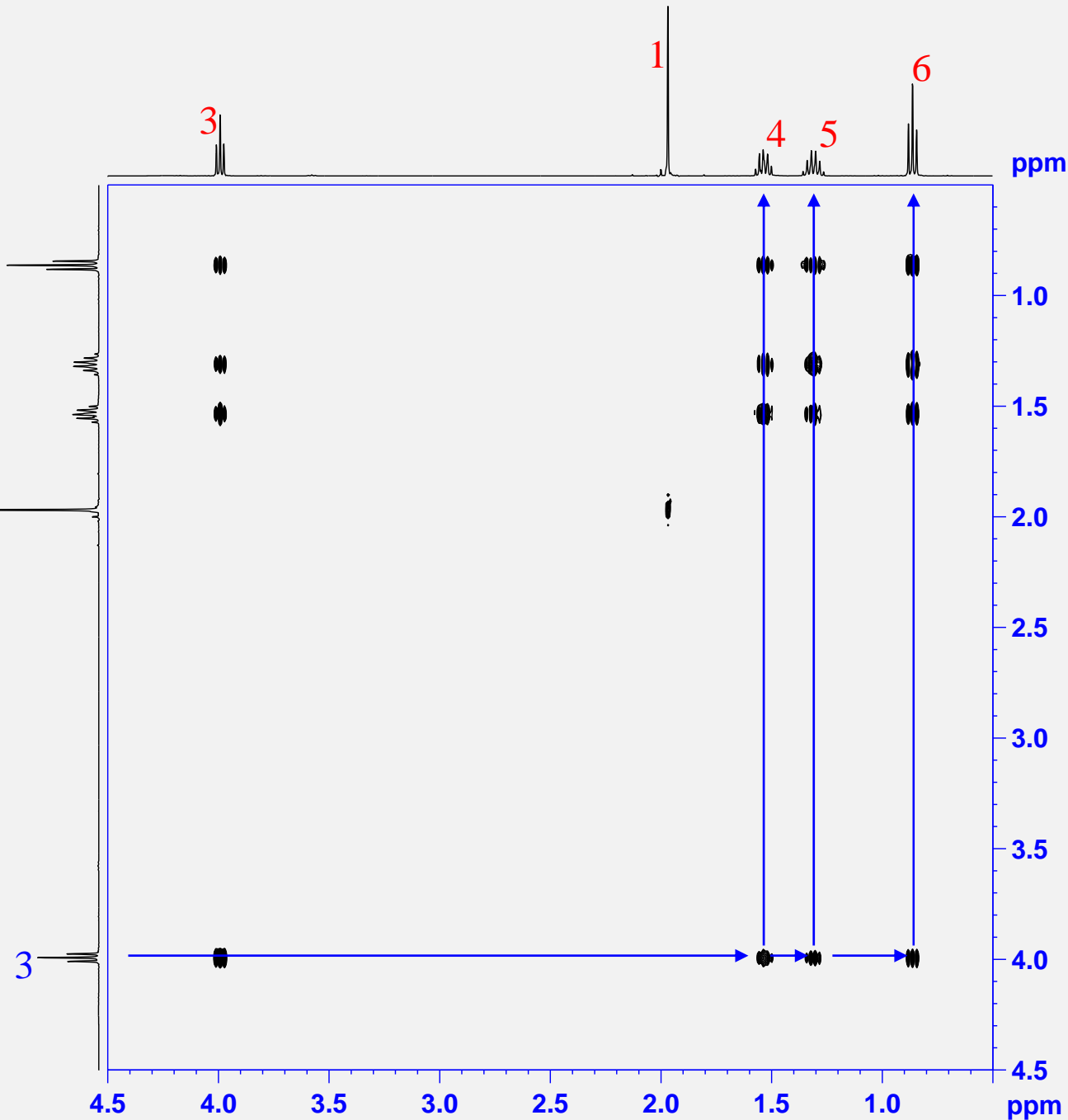


TOCSY

IGRC_2D_TOCSY



n-butyl acetate



3.3 2D homonuclear NOESY/ROESY

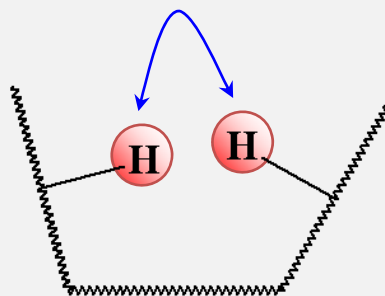
☀ NOESY

(**N**uclear **O**verhauser **E**ffect **S**pectroscopy)

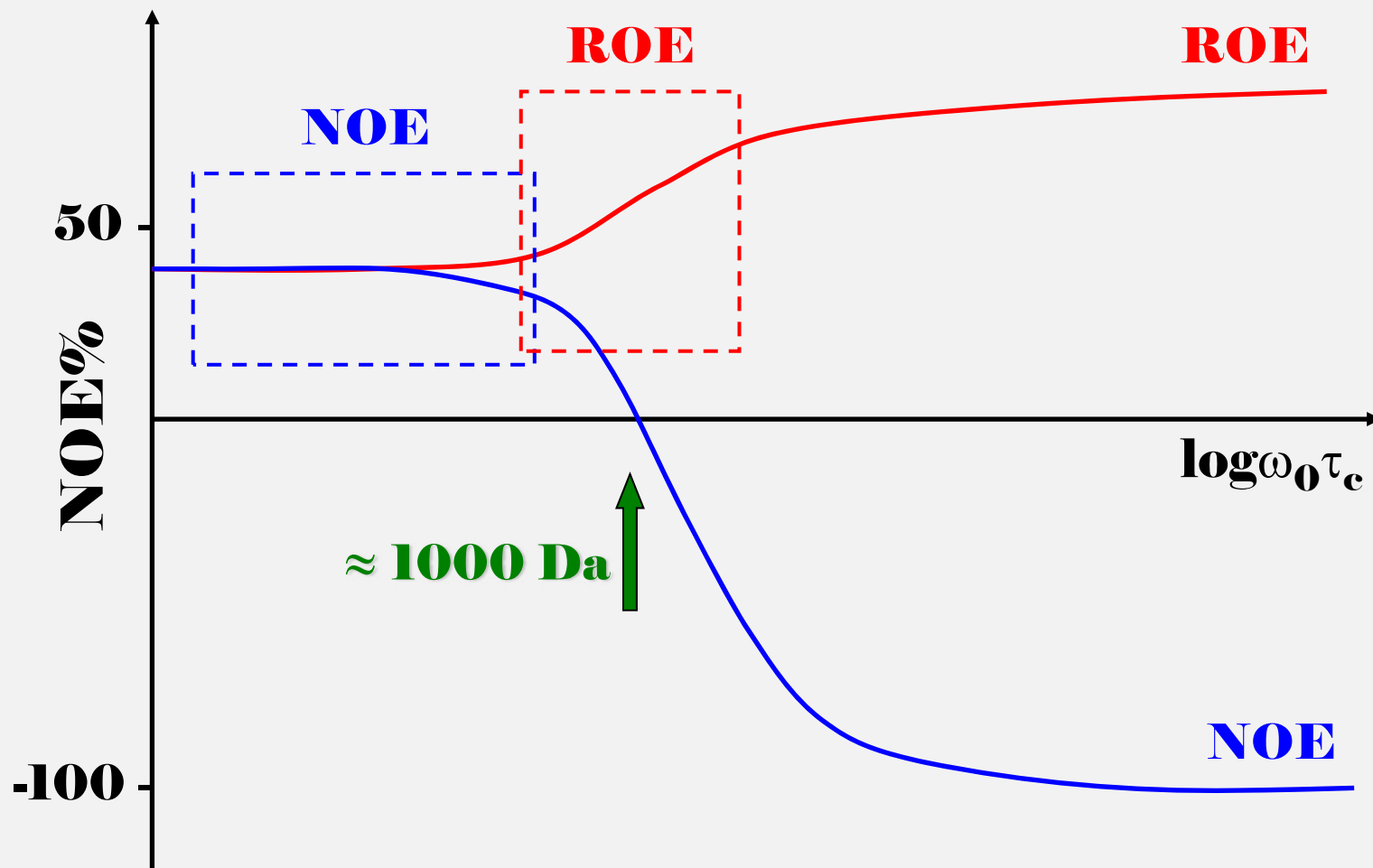
☀ ROESY

(**R**otating-frame **O**verhauser **E**ffect **S**pectroscopy)

獲得空間中之關係, $< 5\text{\AA}$

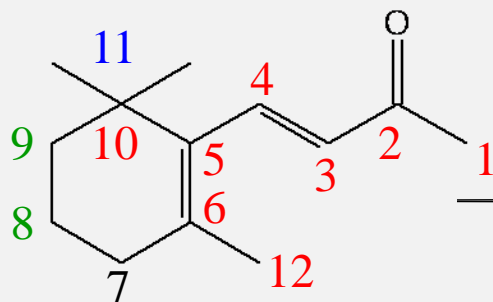


NOESY與ROESY



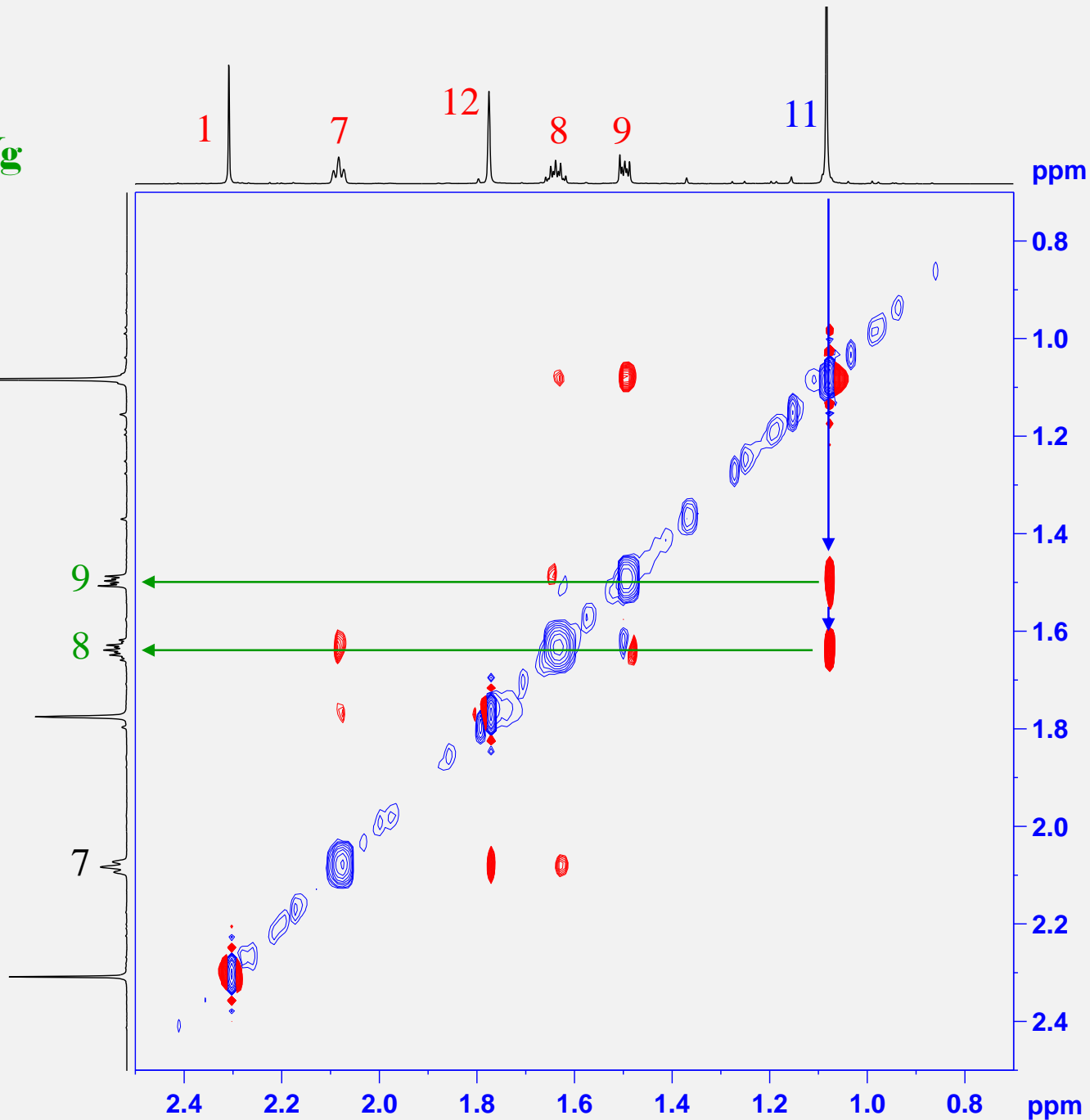
NOESY

IGRC_2D_NOESYg



β -ionone

color of the peak
positive peak
negative peak

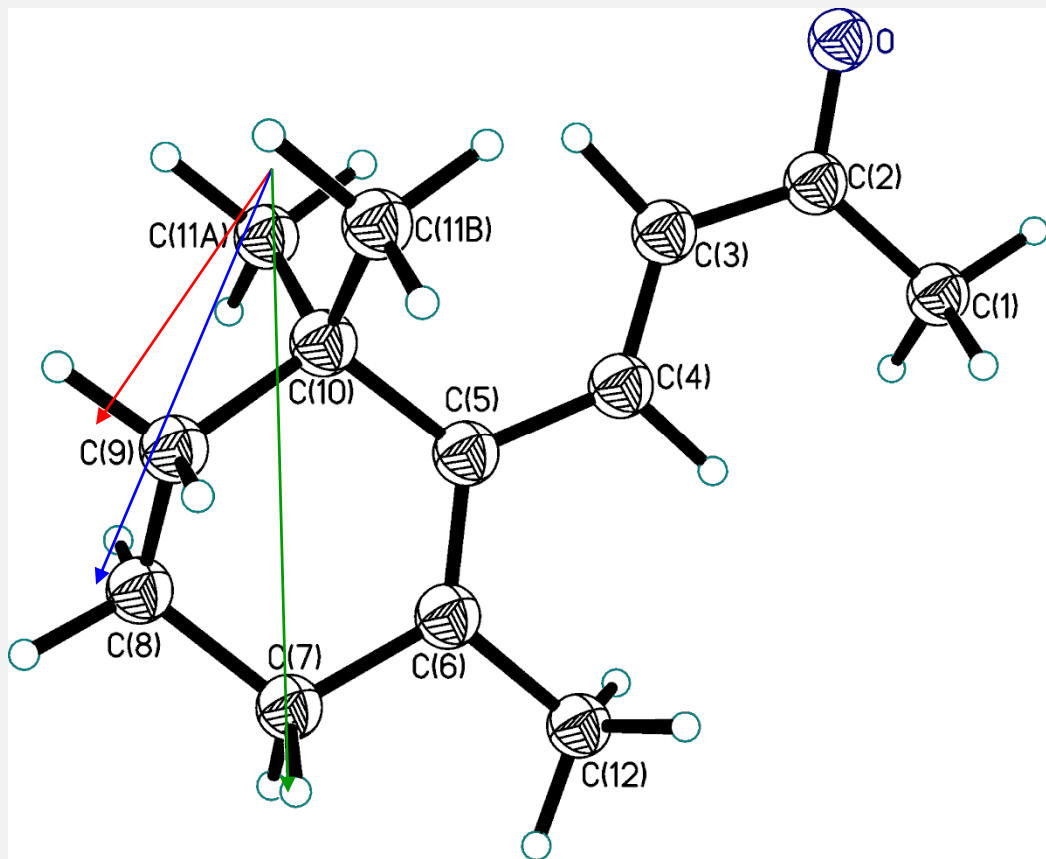


NOESY

**C(11)上的H – C(9)上的H
≈ 2.5Å**

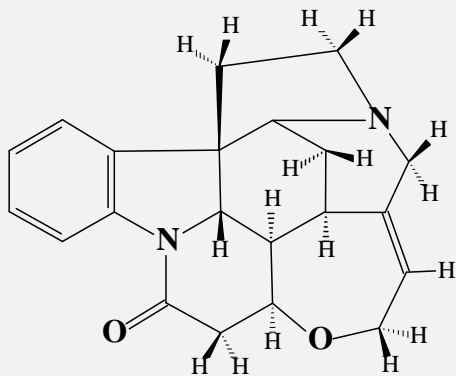
**C(11)上的H – C(8)上的H
≈ 3.8Å**

**C(11)上的H – C(7)上的H
≈ 4.7Å**



ROESY

IGRC_2D_ROESYg

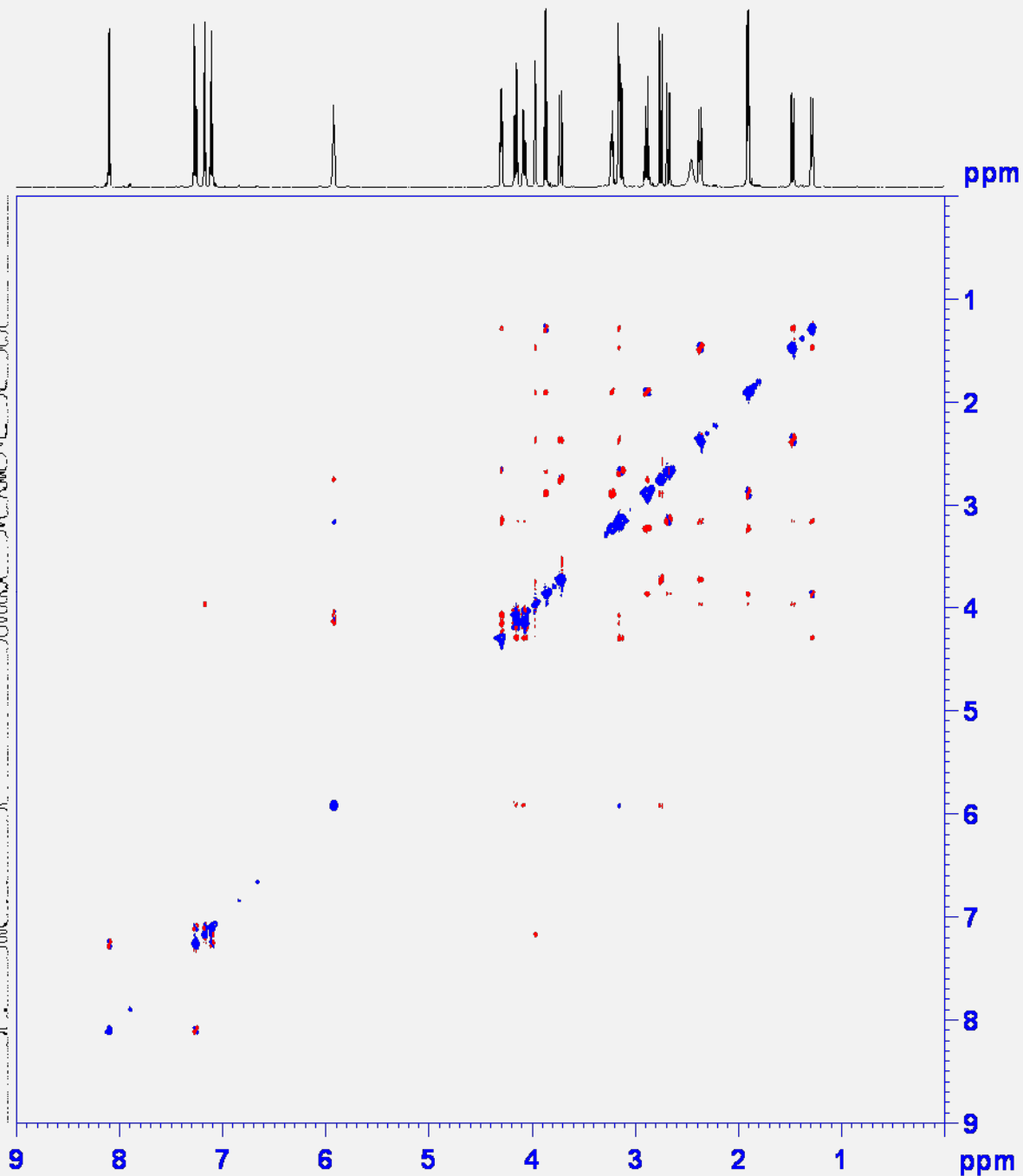


Strychnine

color of the peak

positive peak

negative peak

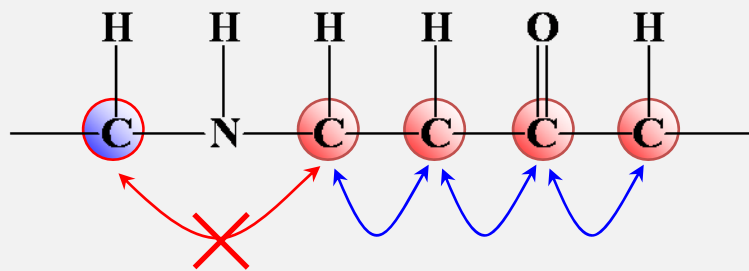


3.4 2D homonuclear ^{13}C

☀ INADEQUATE

(Incredible Natural Abundance Double Quantum Transfer Experiment Spectroscopy)

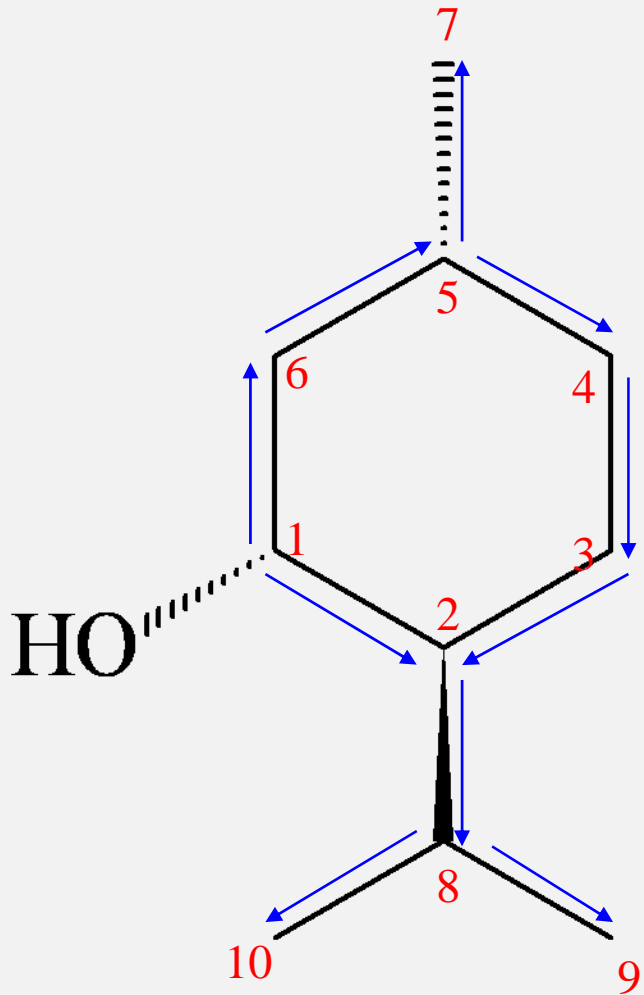
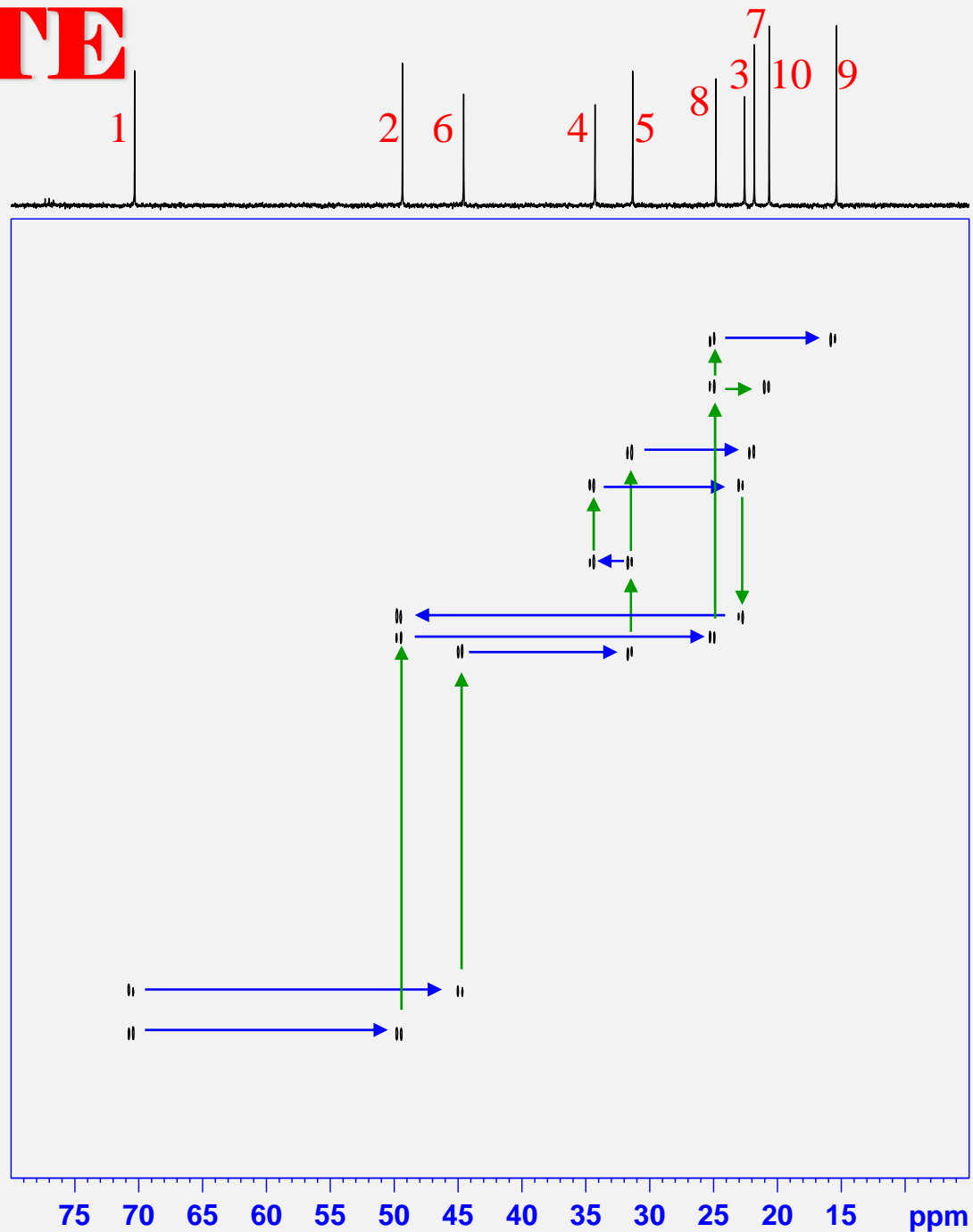
獲得 $^1J_{\text{C-C}}$ 之關係



CryoProbe

INADEQUATE

IGRC_2D_INADEQUATE



4. 2D heteronuclear NMR



HMQC

IGRC_2D_HMQC
IGRC_2D_HMQCs



HMBC

IGRC_2D_HMBC
IGRC_2D_HMBC-suppression



HSQC

IGRC_2D_HSQC
IGRC_2D_HSQC-echo
IGRC_2D_HSQC-suppression



CIGAR-HMBC

IGRC_2D_HMBC-CIGAR



H2BC

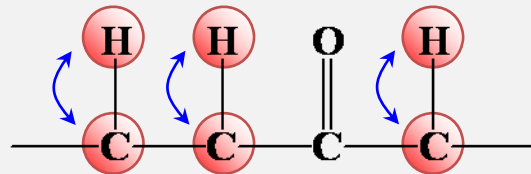
IGRC_2D_H2BC

4.1 2D heteronuclear NMR

☀ HMQC

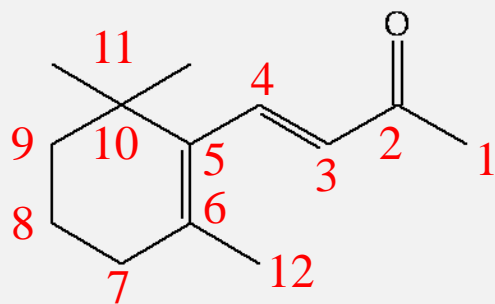
(Heteronuclear Multiple Quantum Correlation)

獲得 $^1J_{\text{H-X}}$ 之關係

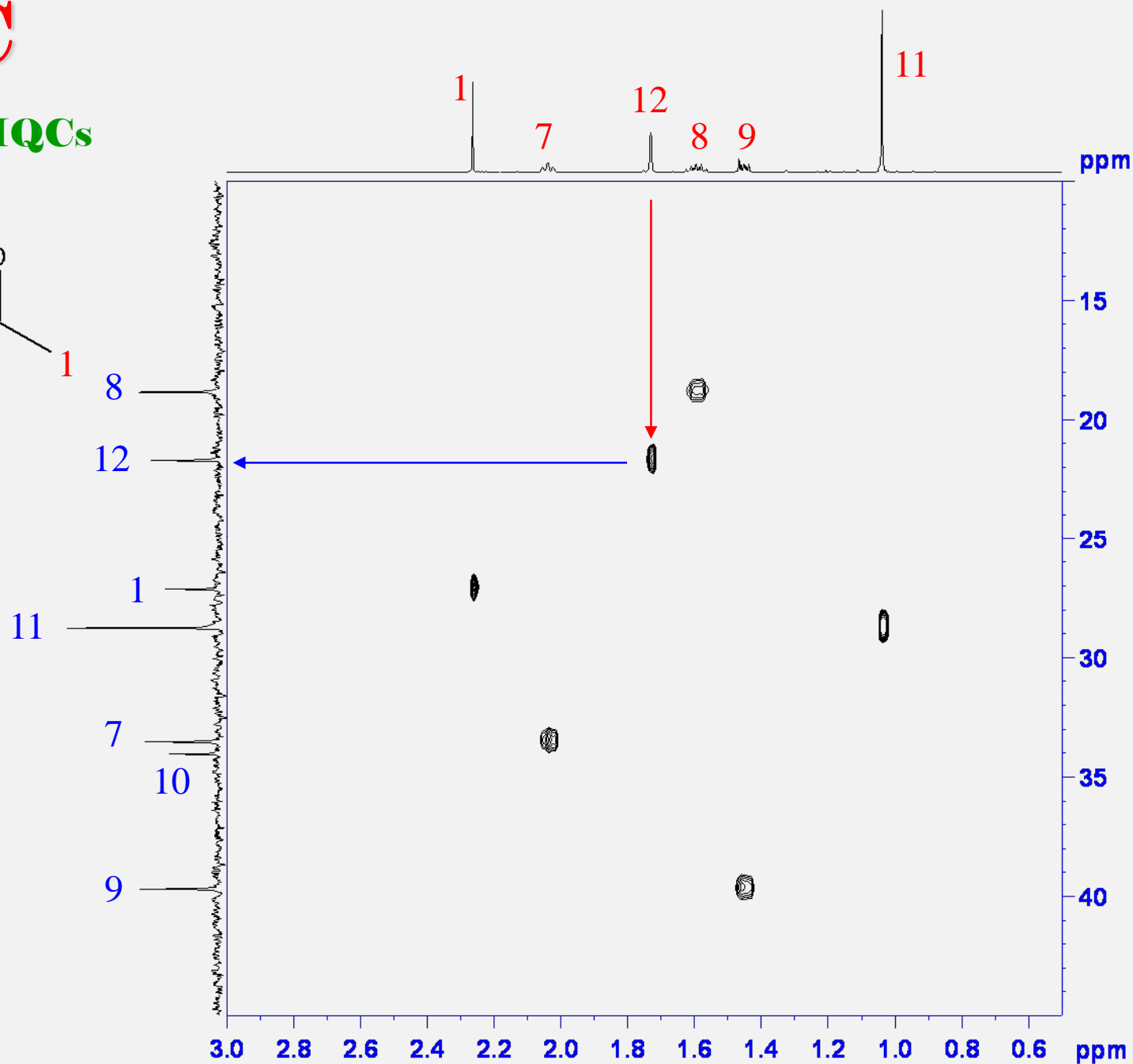


HMQC

IGRC_2D_HMQCs



β -ionone



4.2 2D heteronuclear NMR

☀ HSQC

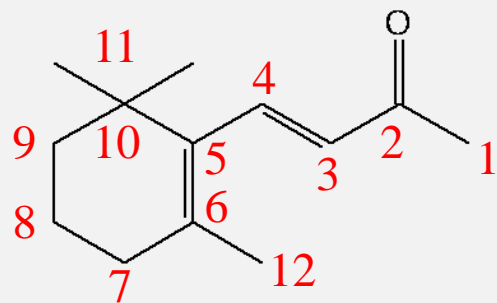
(Heteronuclear Single Quantum Correlation)

獲得 $^1J_{\text{H-X}}$ 之關係

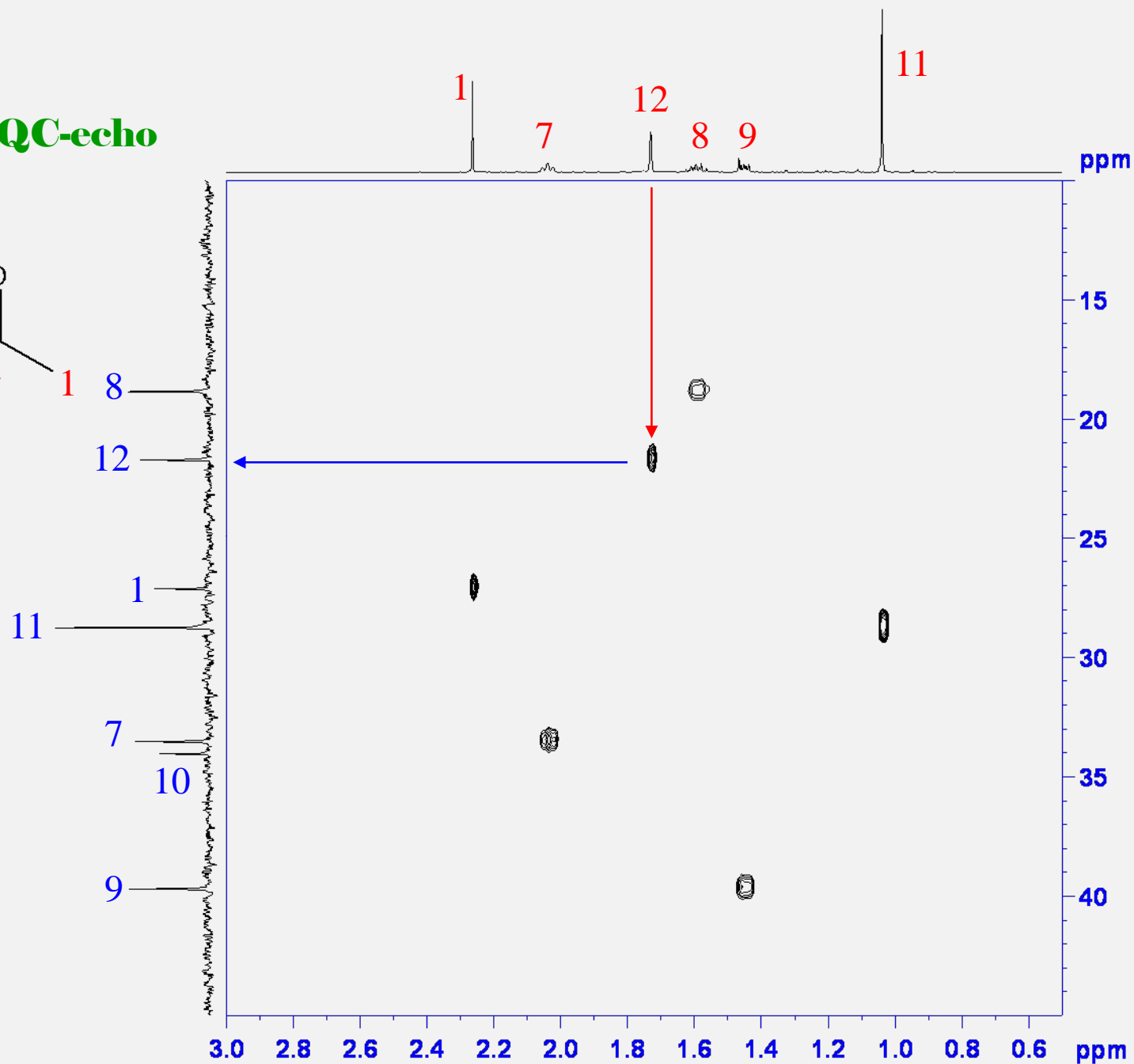


HSQC

IGRC_2D_HSQC-echo



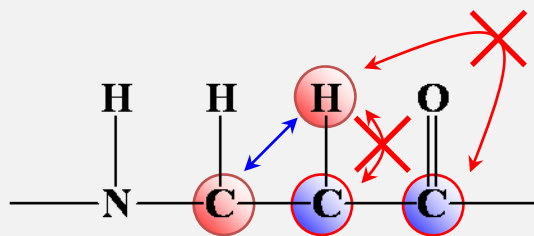
β -ionone



4.3 2D heteronuclear NMR

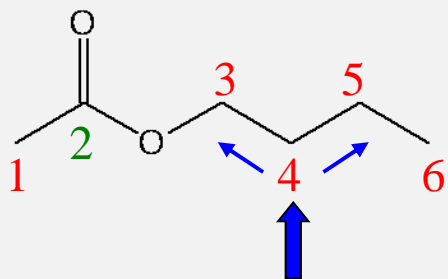
☀ H2BC

獲得 ${}^2J_{\text{H-XH}}$ 之關係



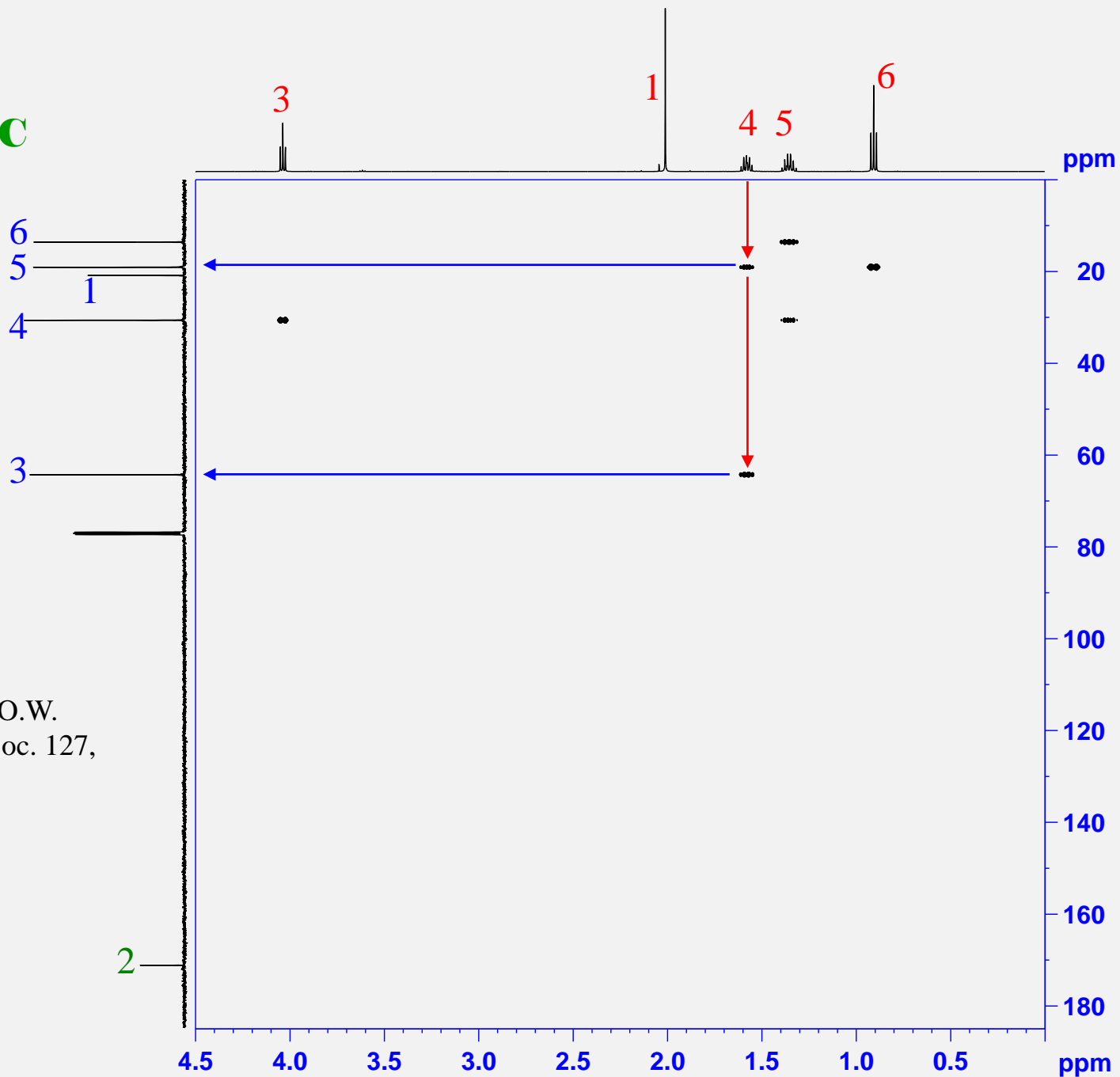
H2BC

IGRC_2D_H2BC



n-butyl acetate

N.T. Nyberg, J.O. Duus & O.W.
Soerensen, J. Am. Chem. Soc. 127,
6154-6155 (2005)



4.4 2D heteronuclear NMR

☀ HMBC

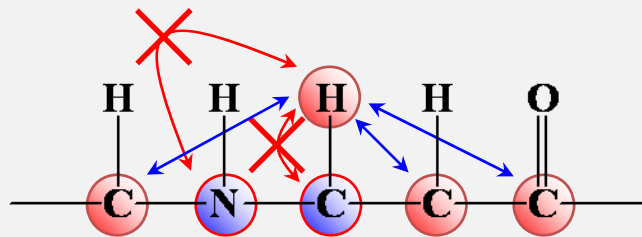
(**H**eteronuclear **M**ultiple **B**ond **C**orrelation)

獲得 ${}^nJ_{\text{H-X}}$ 之關係 ($n \geq 2$)

☀ CIGAR-HMBC

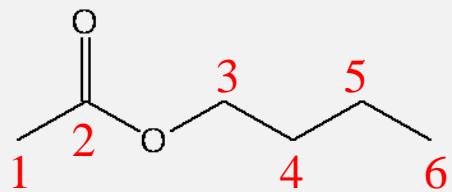
(**C**onstant Time **I**nverse-detected **G**radient **A**ccordion **R**escaled-**H**MBC)

獲得 ${}^nJ_{\text{H-X}}$ 之關係 ($n \geq 2$)，可調整偵測長距離之**H-X**偶合常數

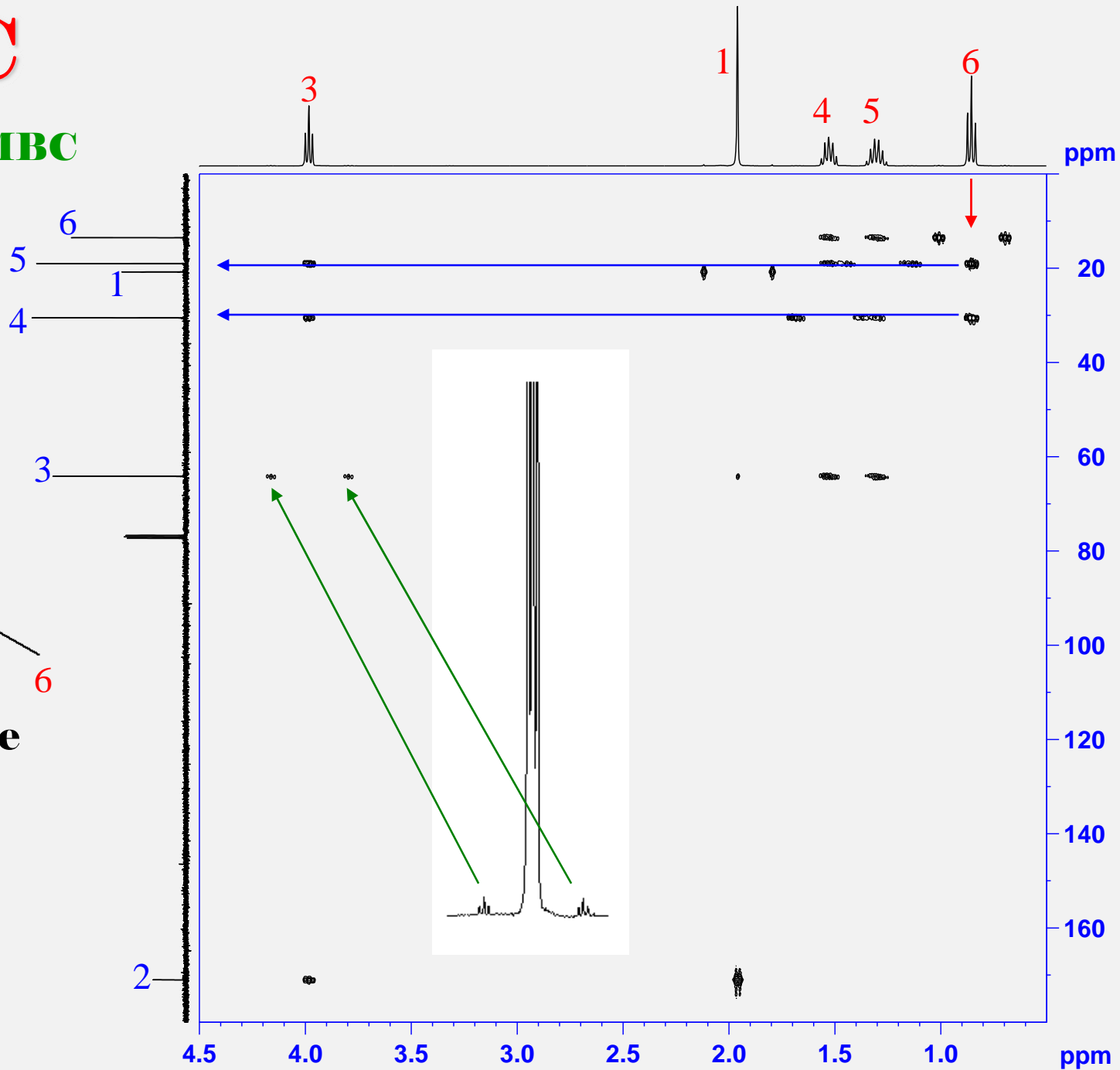


HMBC

IGRC_2D_HMBC

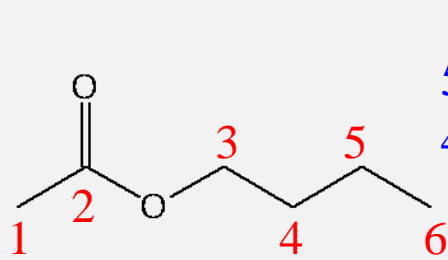


n-butyl acetate



CIGAR-HMBC

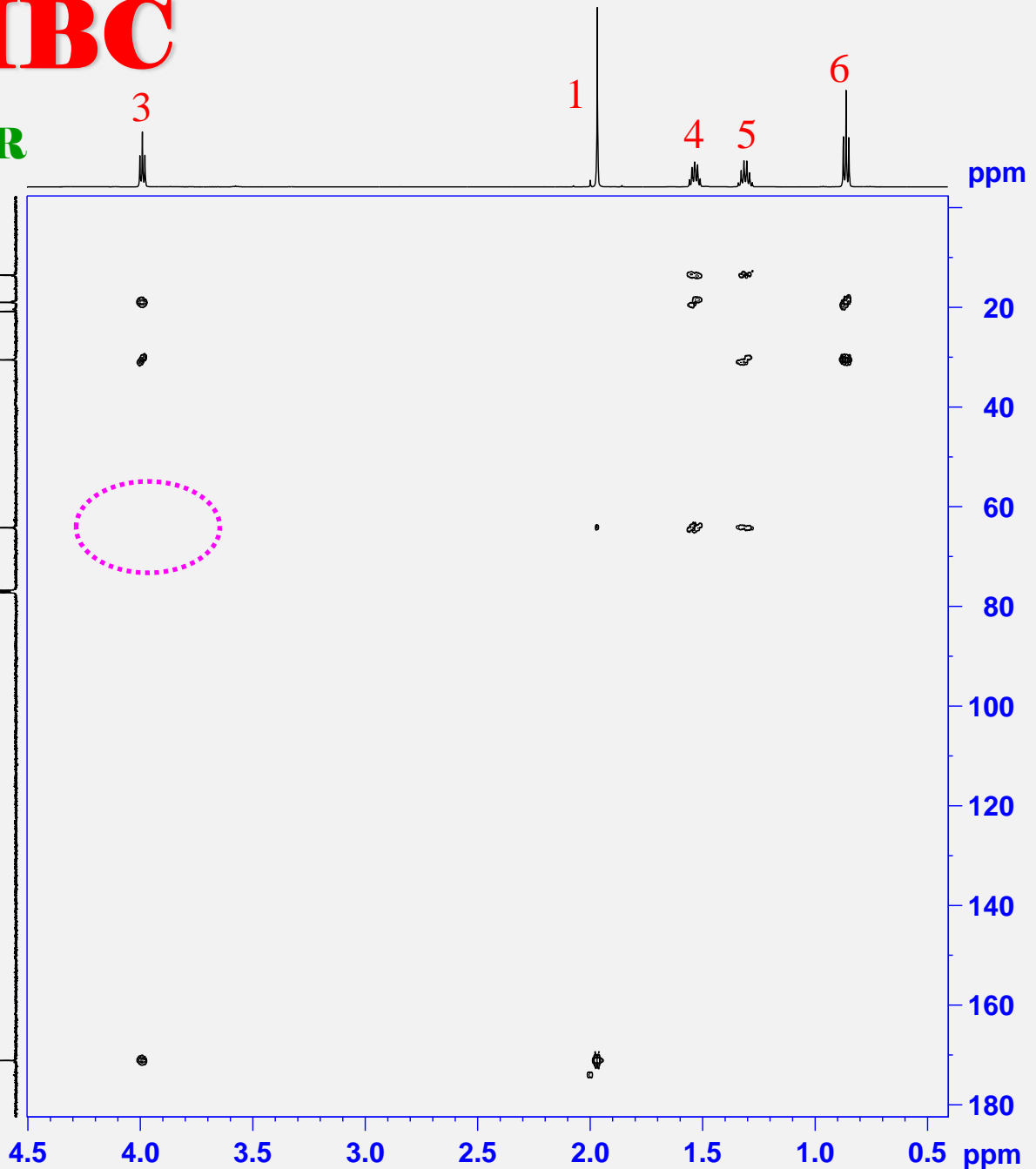
IGRC_2D_HMBC-CIGAR



n-butyl acetate

1. V.V. Krishnamurthy, D.J. Russel, C.E. Hadden & G.E. Martin, J. Magn. Reson. 146, 232-239 (2000)
2. C.E. Hadden, G.E. Martin & V.V. Krishnamurthy, Magn. Reson. Chem. 38, 143-147 (2000)

1. 訊號強度較HMBC 少4倍
2. 可去除satellite 訊號



5. Measurement of coupling constants

Homonuclear J-resolved

1GRC_2D_Jres

Heteronuclear J-resolved

1GRC_2D_HJres

1GRC_2D_sel.HJres

1GRC_2D_no decouple HSQC

1GRC_2D_HETLOC

1GRC_2D_HSQC-HECADE

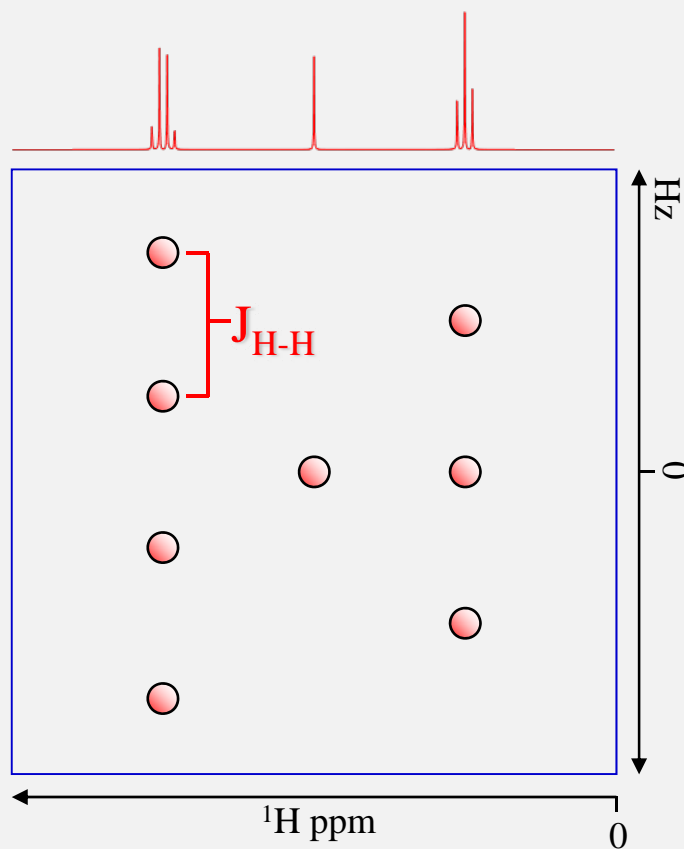
1GRC_2D_ps-HMBC

1GRC_2D_G-BIRD-HSQMBC

5.1 Measurement of coupling constants

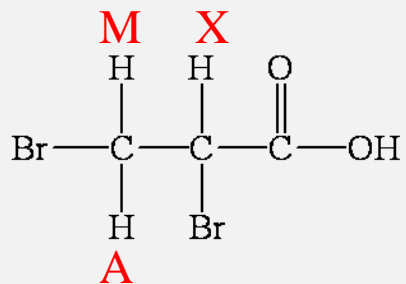
☀ homonuclear J-resolved

獲得 J_{H-H} 之偶合常數

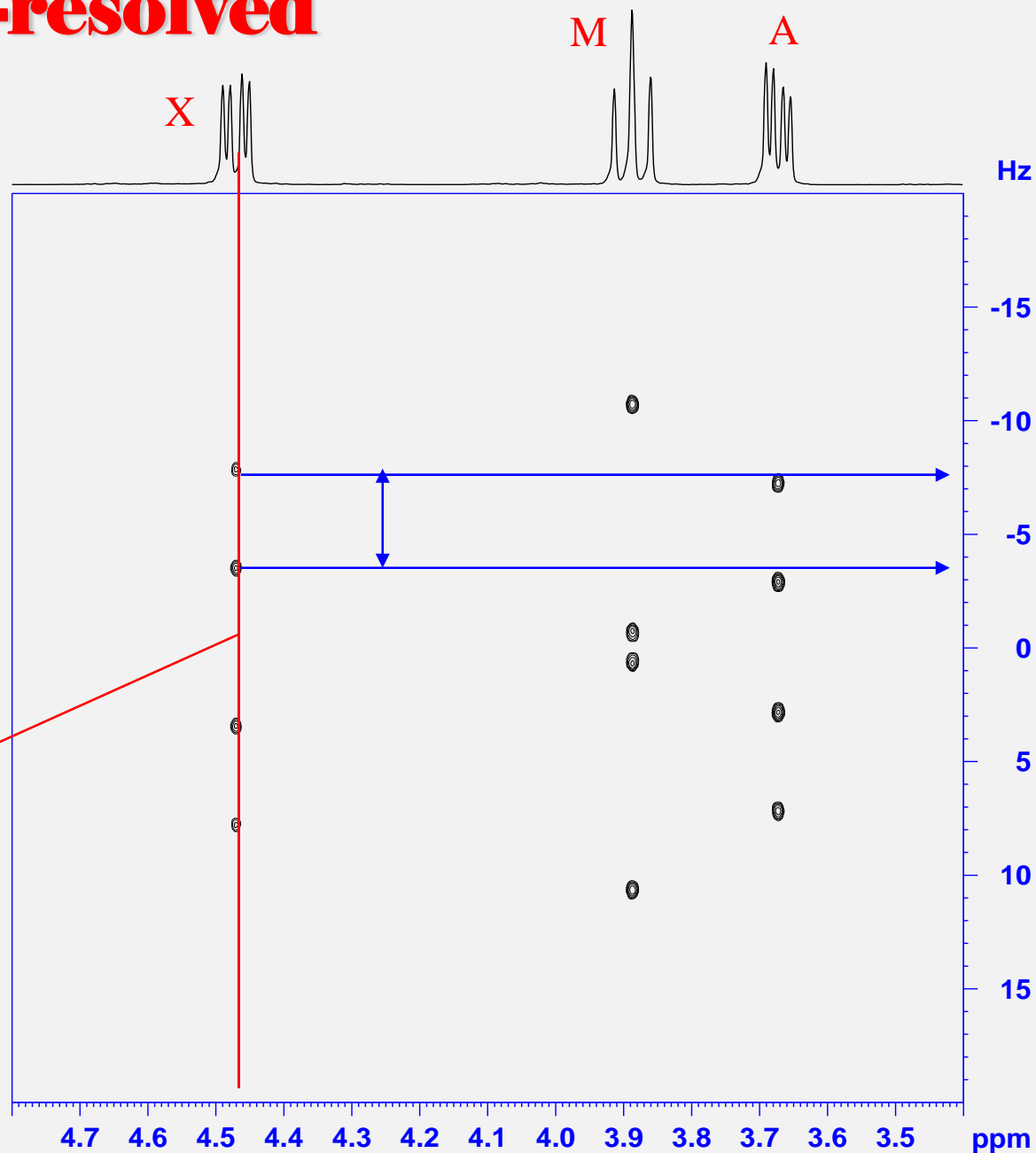
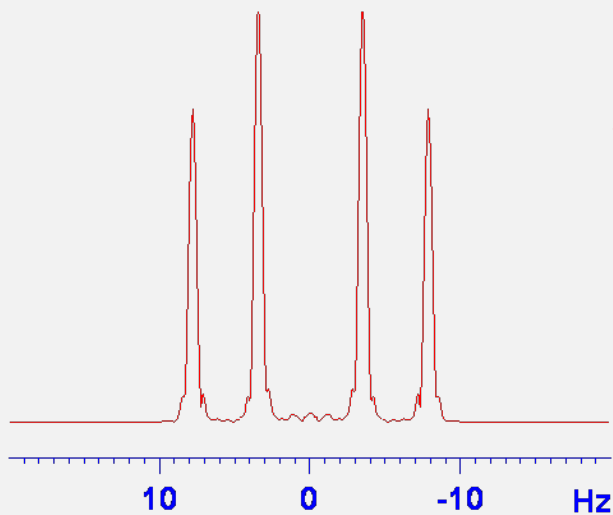


homonuclear J-resolved

IGRC_2D_Jres



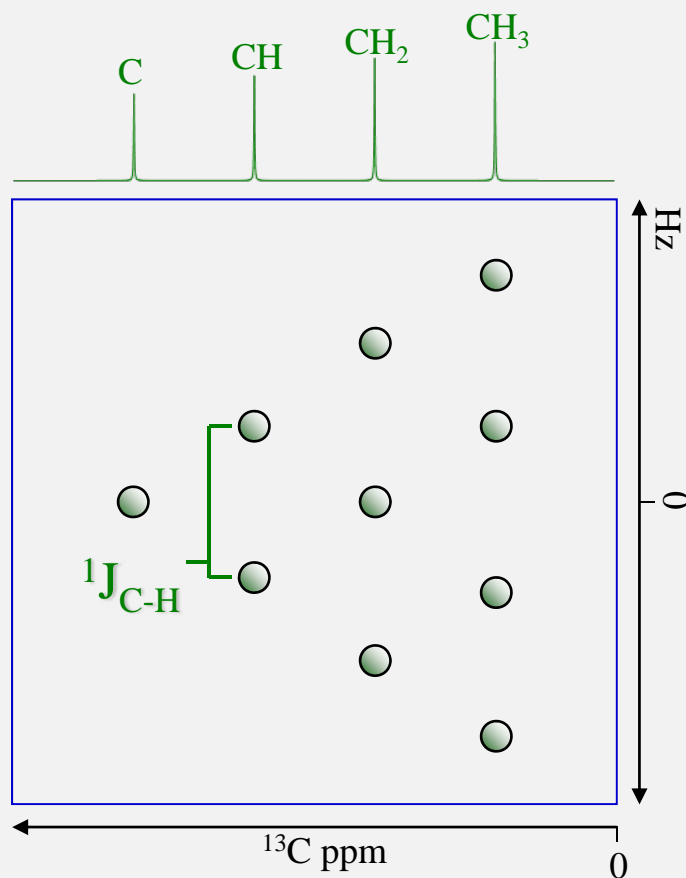
**2,3-dibromo-
propionic acid**



5.2 Measurement of coupling constants

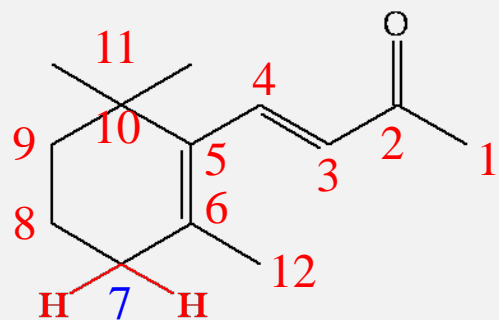
☀ heteronuclear J-resolved

獲得 J_{C-H} 之耦合常數

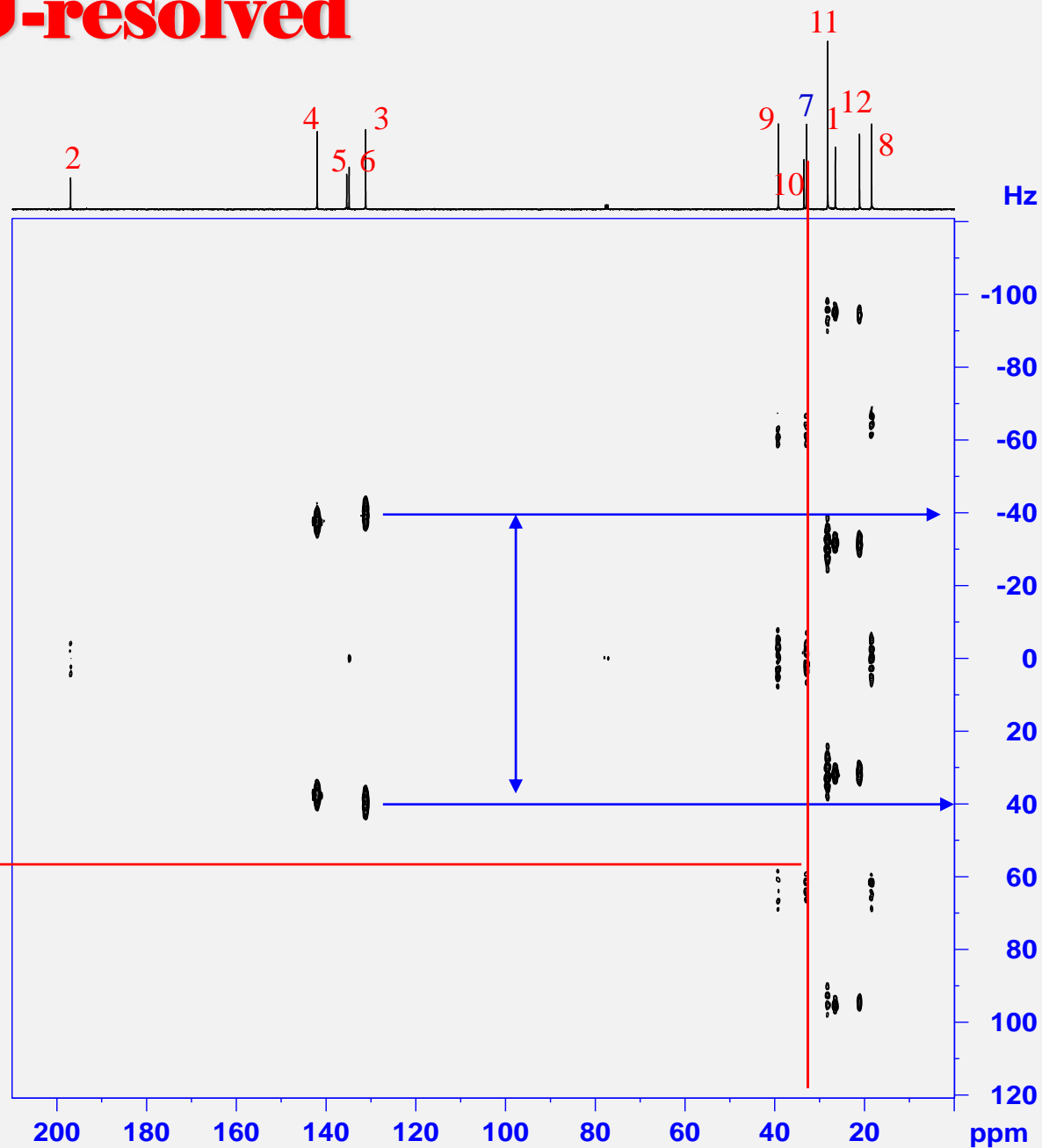
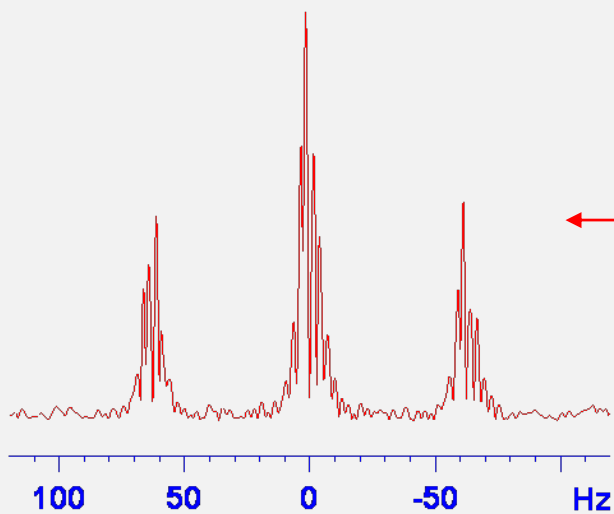


heteronuclear J-resolved

IGRC_2D_HJres

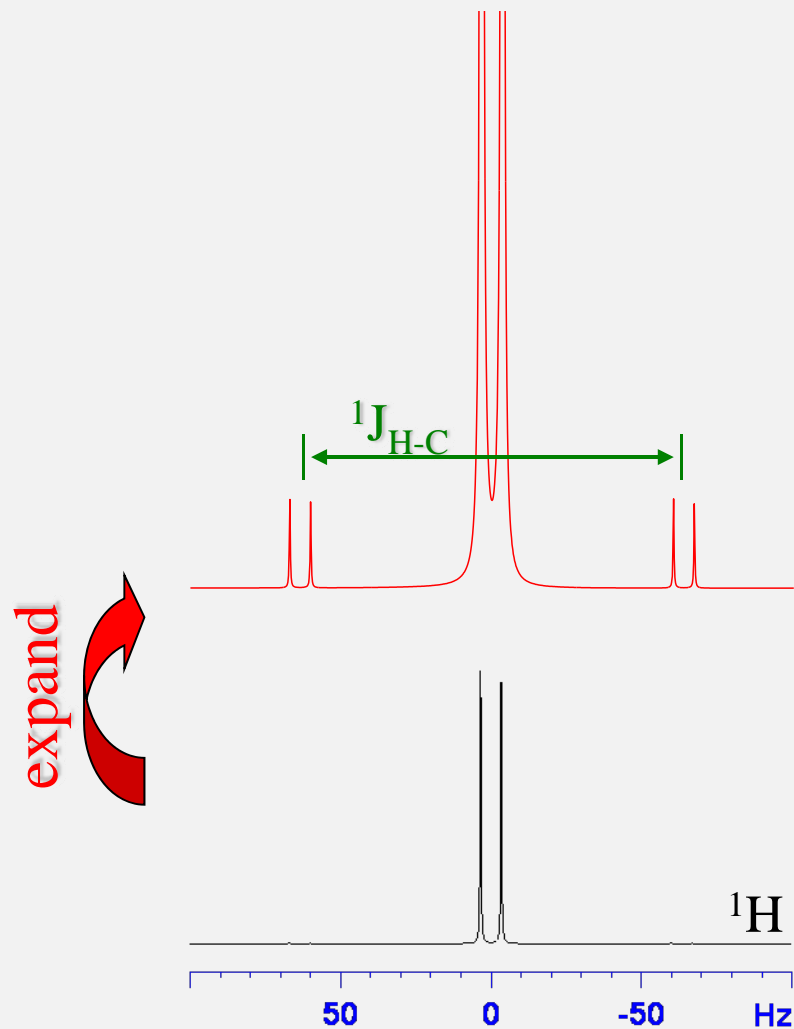
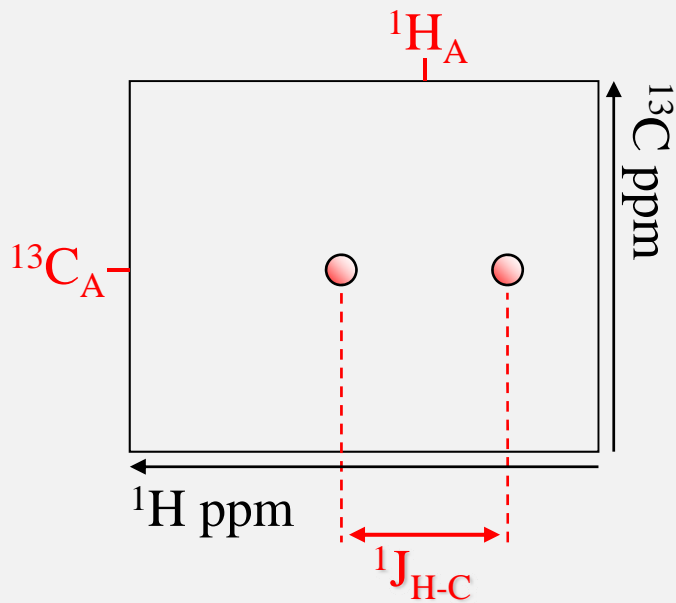


β -ionone



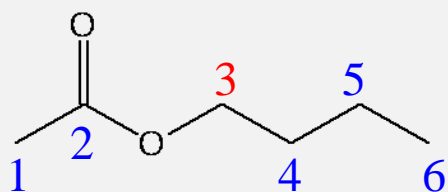
5.3 Measurement of coupling constants

☀ **no decoupling HSQC**
獲得 $^1J_{\text{H-HC}}$ 之耦合常數

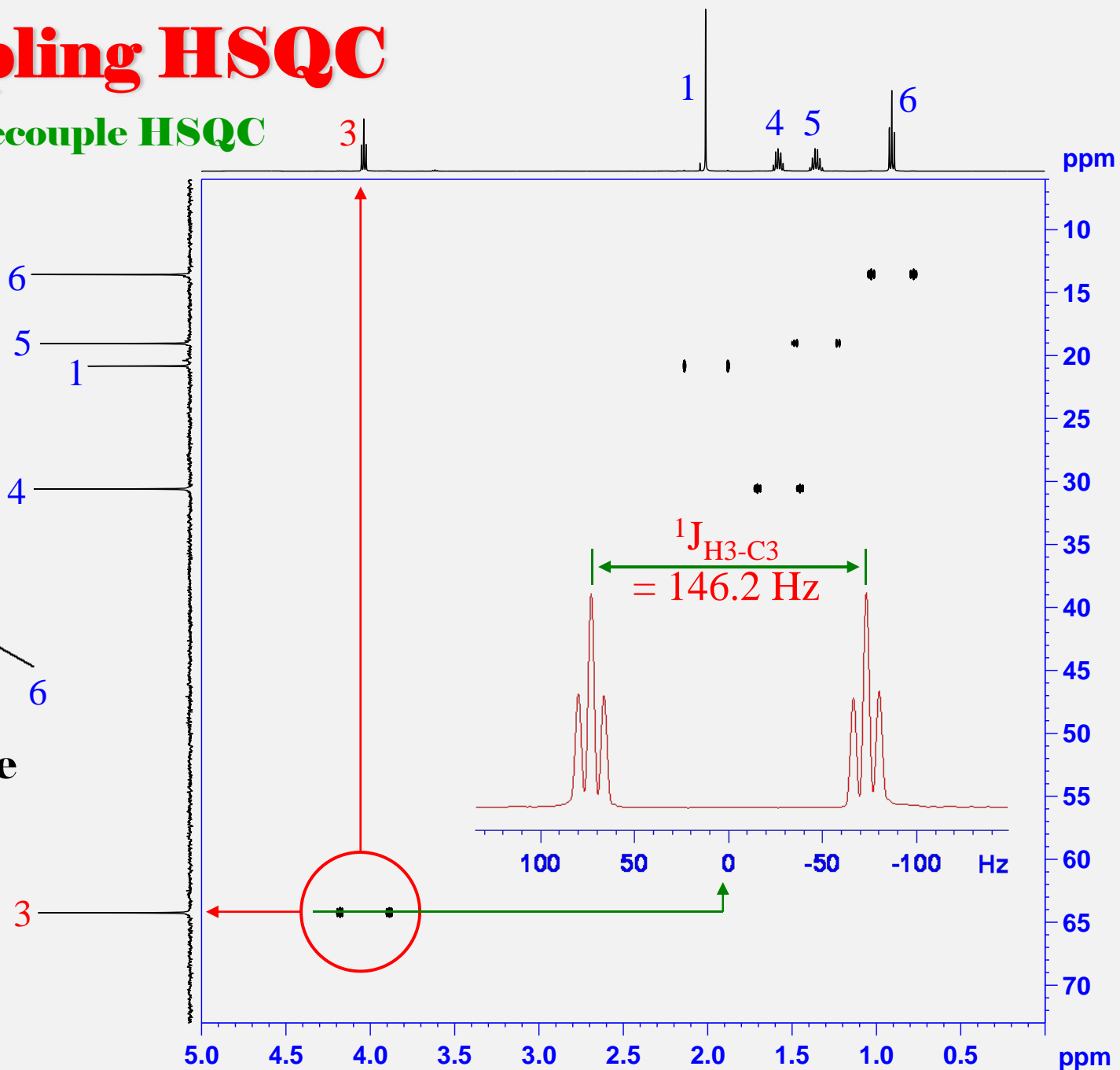


no decoupling HSQC

IGRC_2D_no decouple HSQC



n-butyl acetate

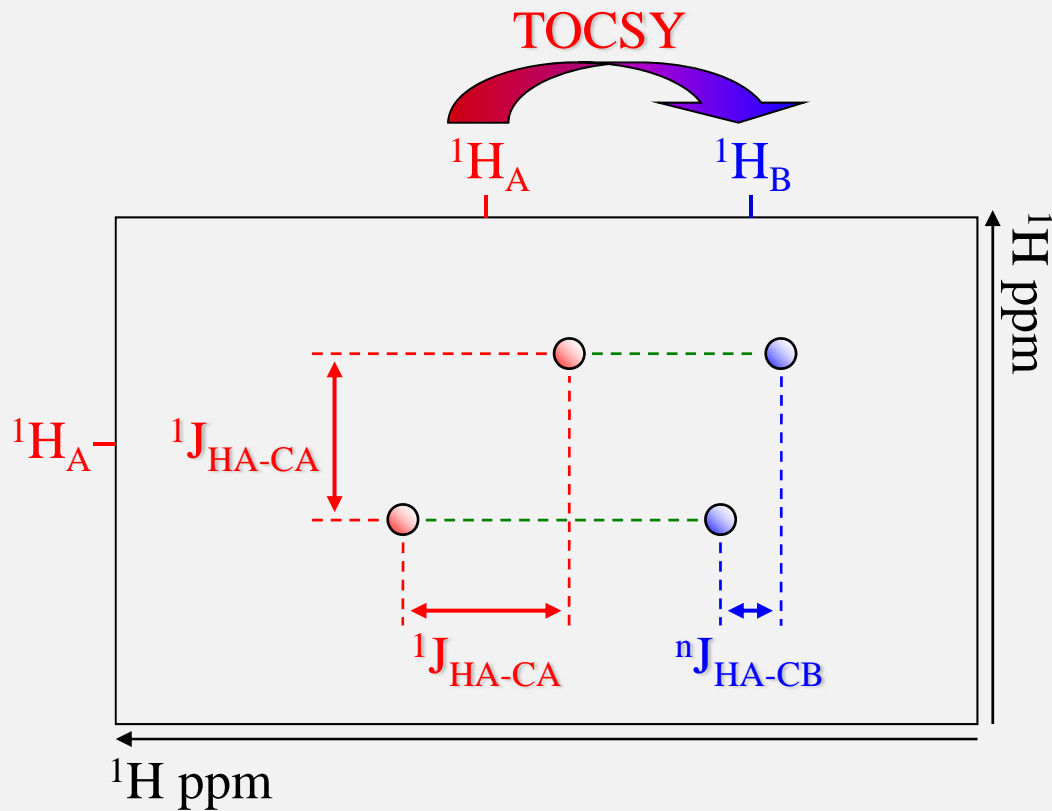


5.4 Measurement of coupling constants

☀ HETLOC

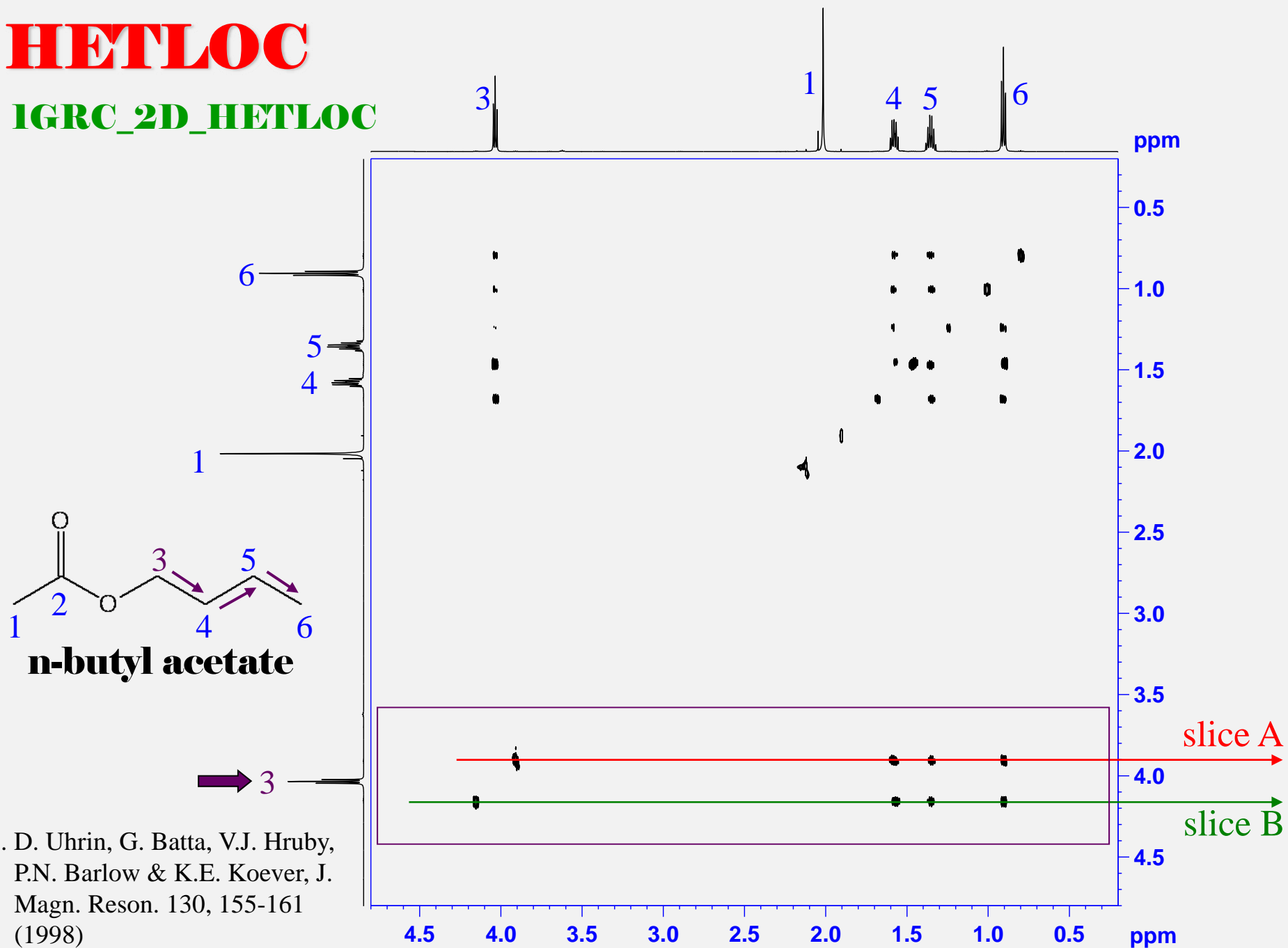
(HETeronuclear LOng-range Couplings)

獲得 $nJ_{\text{H-HC}}$ 之偶合常數($n \geq 1$)



HETLOC

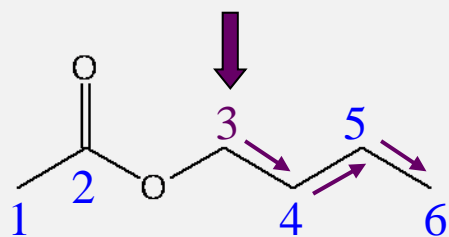
IGRC_2D_HETLOC



1. D. Uhrin, G. Batta, V.J. Hruby,
P.N. Barlow & K.E. Koeber, J.
Magn. Reson. 130, 155-161
(1998)

HETLOC

IGRC_2D_HETLOC



n-butyl acetate

H3
 $^1J_{\text{H3-C3}}$
= 146.2 Hz

H4
 $^2J_{\text{H3-C4}}$
= 4.4 Hz

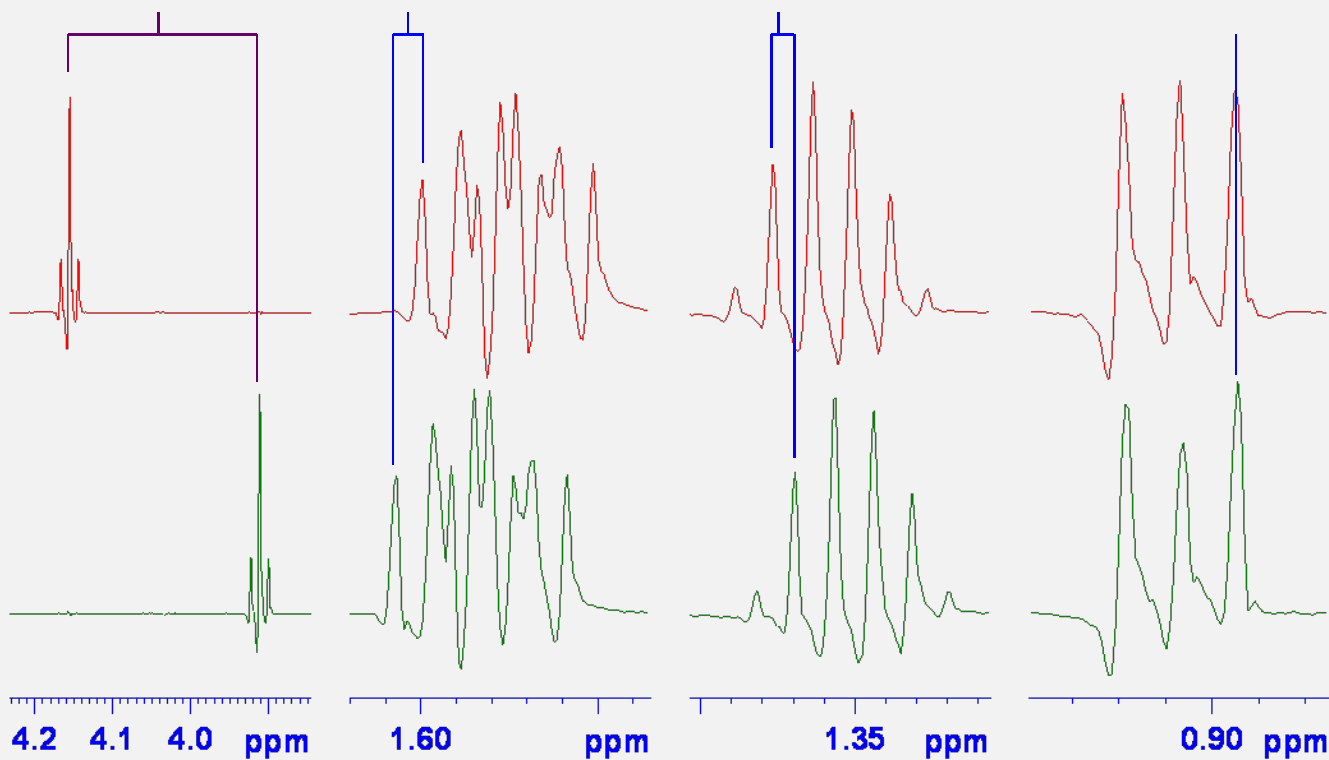
H5
 $^3J_{\text{H3-C5}}$
= 4.1 Hz

H6
 $^4J_{\text{H3-C6}}$
 ≈ 0 Hz

slice A

H3

slice B

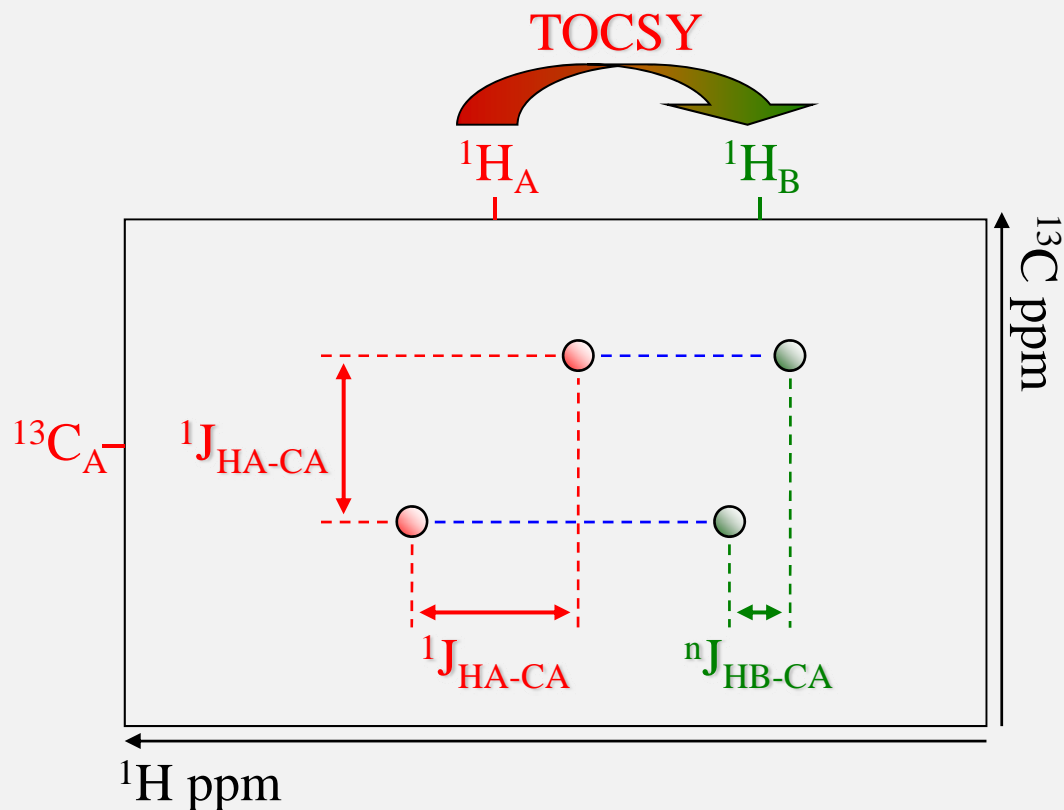


5.5 Measurement of coupling constants

☀ HSQC-HECADE

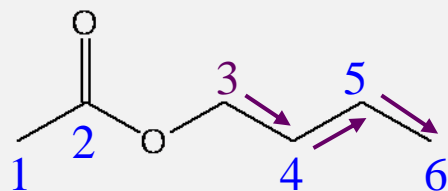
(HSQC-HEteronuclear C Couplings from ASSCI-Domain Experiments with E.COSY-Type Cross Peaks)

獲得 $nJ_{\text{H-HC}}$ 之偶合常數 ($n \geq 1$)



HSQC-HECADE

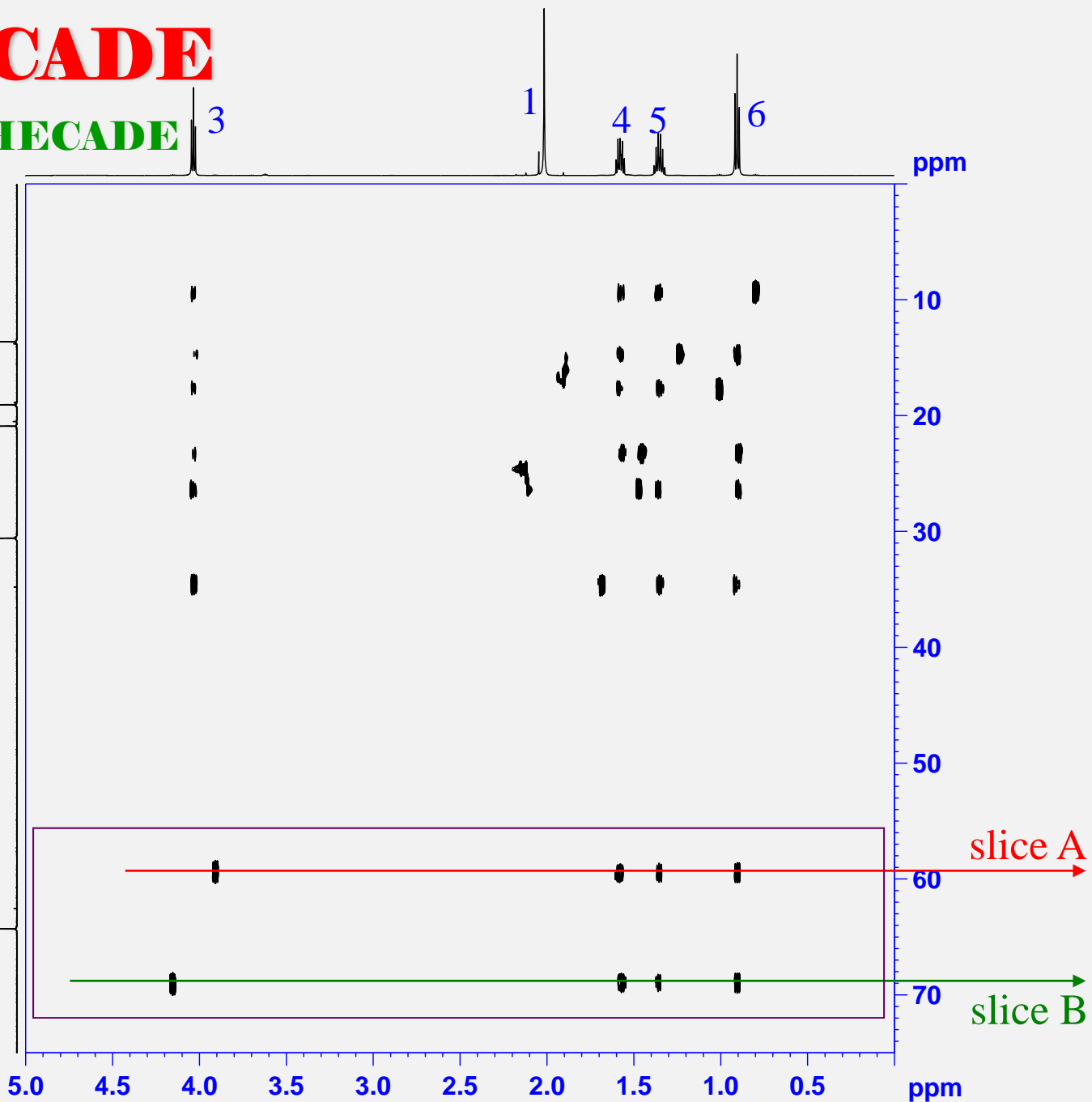
IGRC_2D_HSQC-HECADE



n-butyl acetate

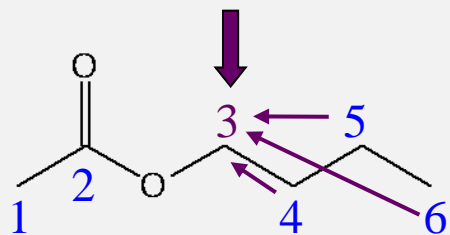


W. Kozminski & D. Nanz, J.
Magn. Reson. 142, 294-299
(2000)



HSQC-HECADE

IGRC_2D_HSQC-HECADE



n-butyl acetate

H3 \rightarrow H4 \rightarrow H5 \rightarrow H6

$^1J_{\text{H3-C3}} = 146.2 \text{ Hz}$

$^2J_{\text{H4-C3}} = 4.6 \text{ Hz}$

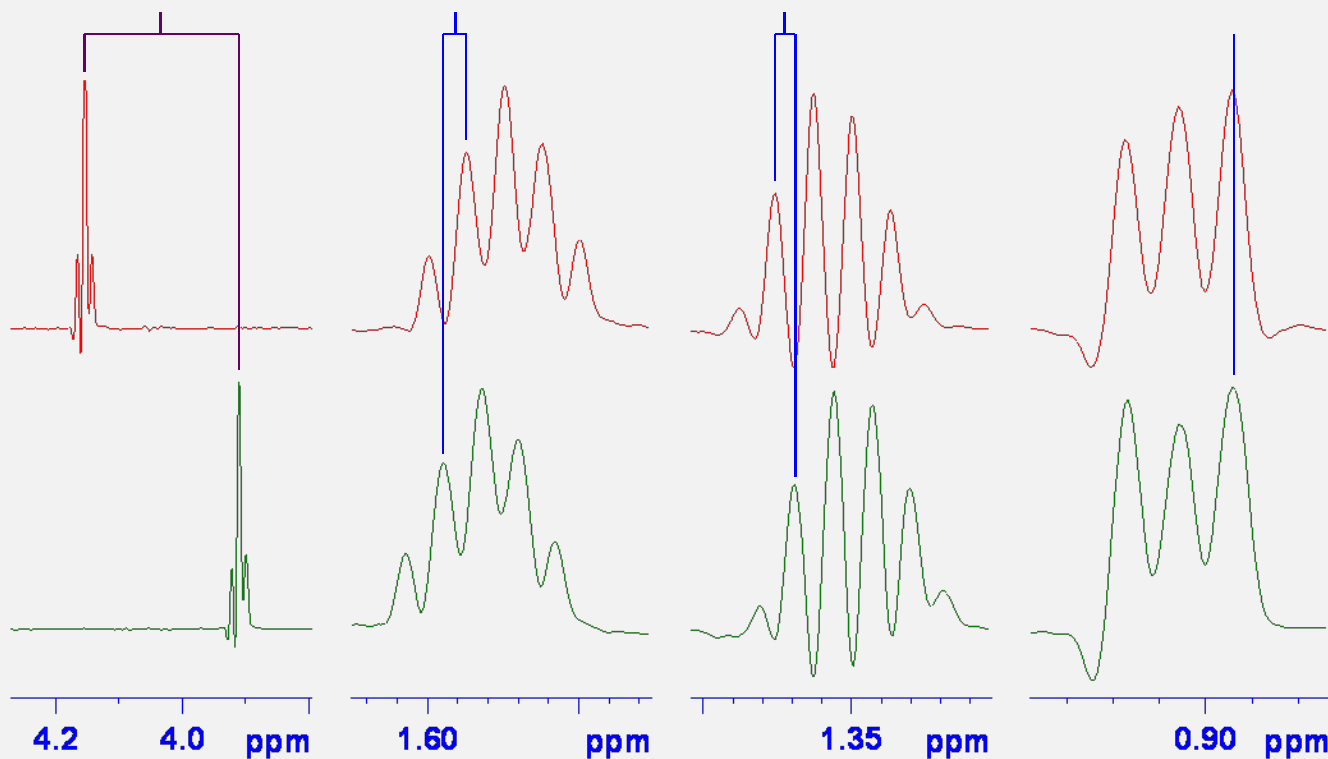
$^3J_{\text{H5-C3}} = 4.1 \text{ Hz}$

$^4J_{\text{H6-C3}} \approx 0 \text{ Hz}$

slice A \rightarrow

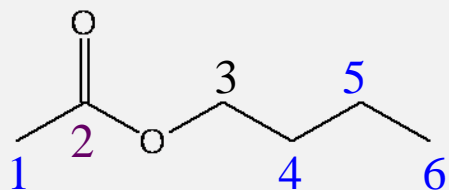
\rightarrow C3

slice B \rightarrow



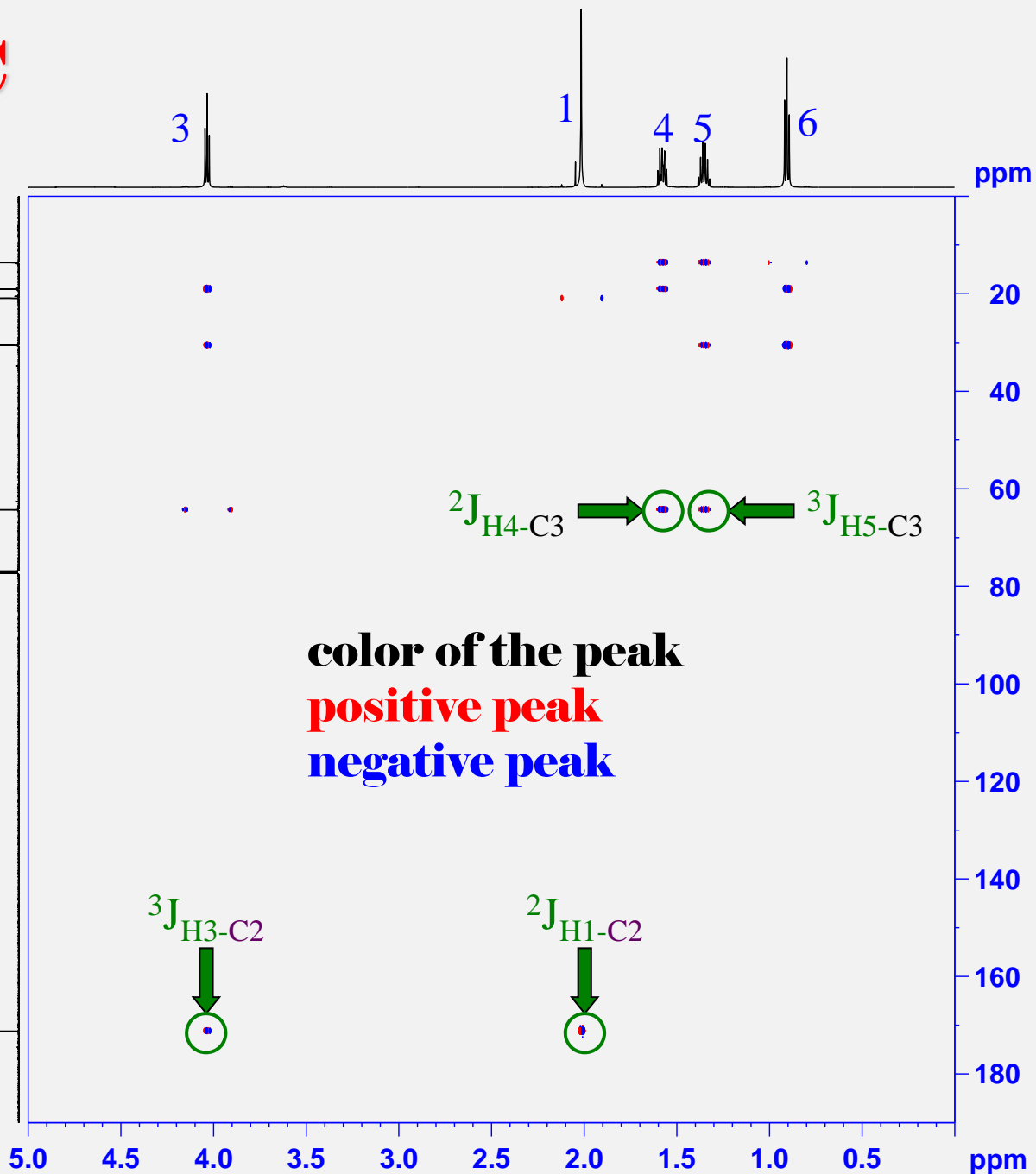
5.6 ps-HMBC

1GRC_2D_ps-HMBC



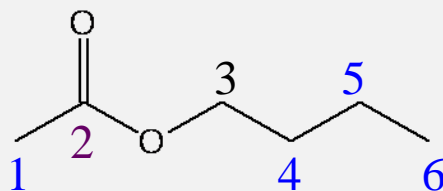
n-butyl acetate

D.O. Cicero, G. Barbato & R. Bazzo, J. Magn. Reson. 148, 209-213 (2001)



ps-HMBC

IGRC_2D_ps-HMBC



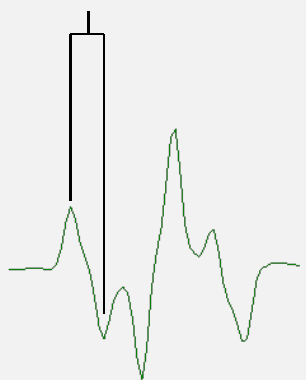
n-butyl acetate

$${}^3J_{\text{H3-C2}} = 3.1 \text{ Hz}$$

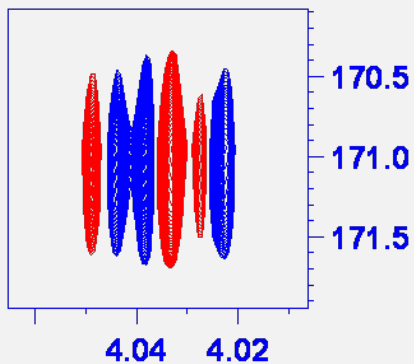
$${}^2J_{\text{H1-C2}} = 6.7 \text{ Hz}$$

$${}^2J_{\text{H4-C3}} = 4.6 \text{ Hz}$$

$${}^3J_{\text{H5-C3}} = 4.2 \text{ Hz}$$

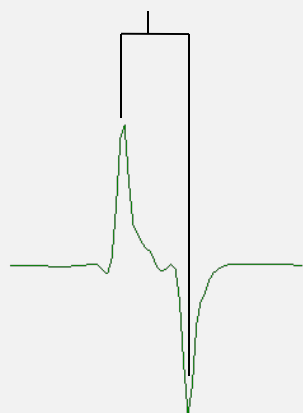


4.04 ppm

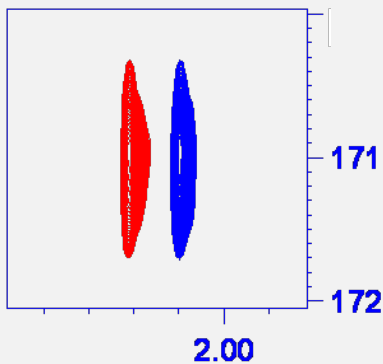


170.5
171.0
171.5

4.04 4.02



2.02 ppm

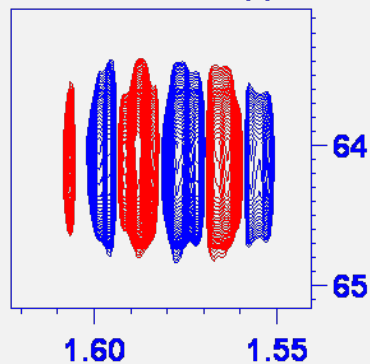


171
172

2.00

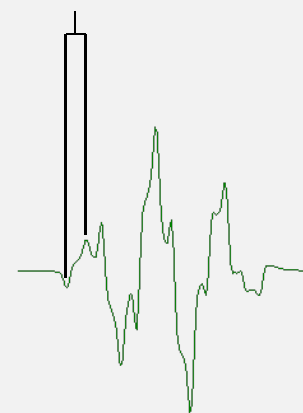


1.60 ppm

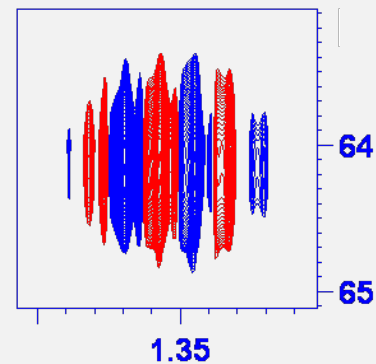


64
65

1.60 1.55



1.35 ppm

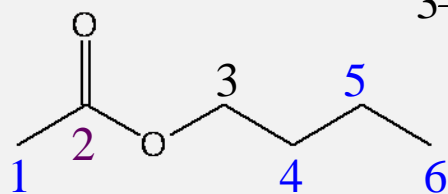


64
65

1.35

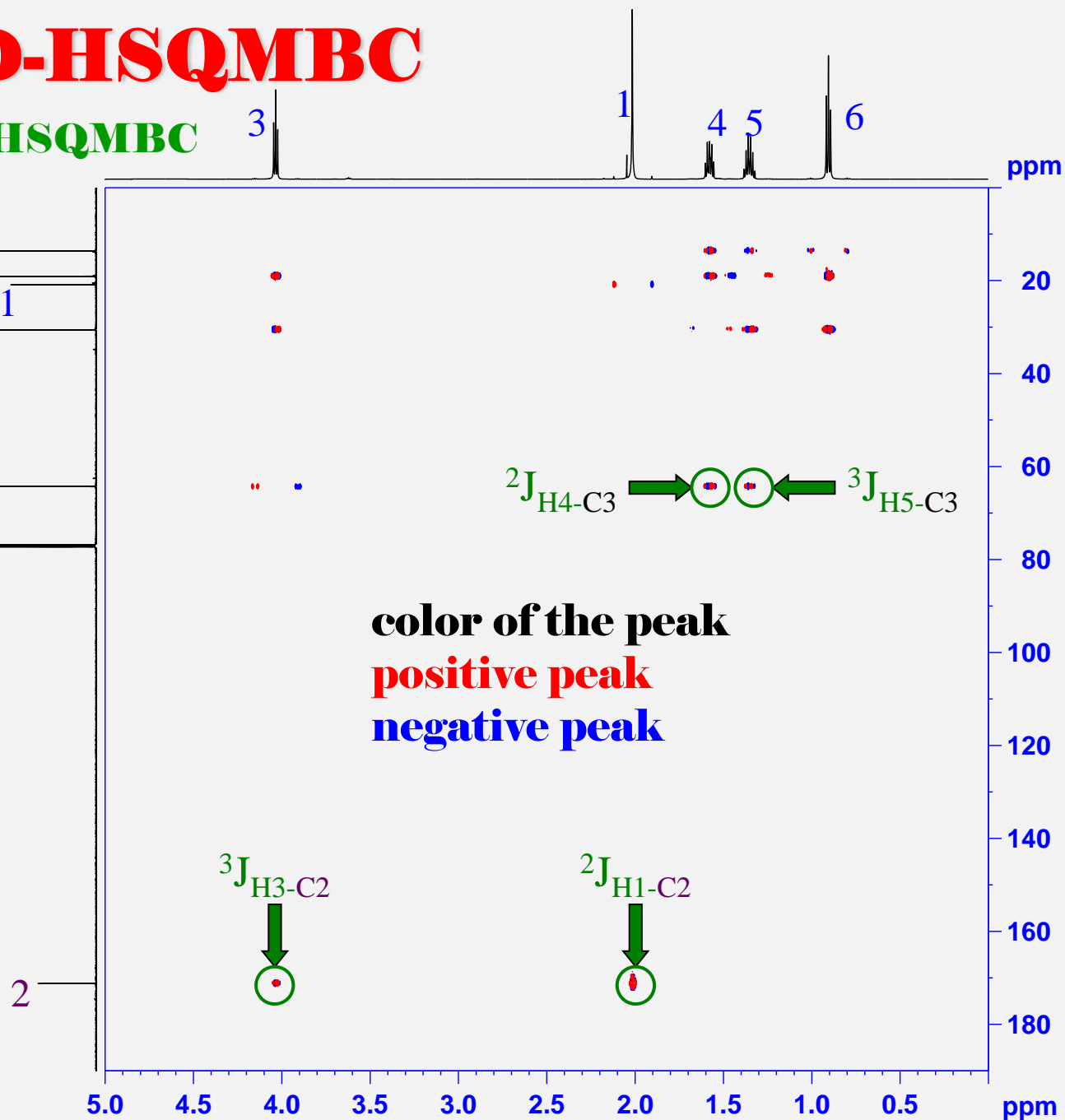
5.7 G-BIRD-HSQMBC

IGRC_2D_G-BIRD-HSQMBC



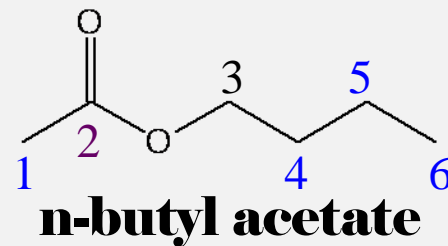
n-butyl acetate

1. B.L. Marquez, W.H. Gerwick & R.T. Williamson, *Magn. Reson. Chem.* 39, 499-530 (2001)
2. R.T. Williamson, B.L. Marquez, W.H. Gerwick & K.E. Kover, *Magn. Reson. Chem.* 38, 265-273 (2000)



G-BIRD-HSQMBC

IGRC_2D_G-BIRD-HSQMBC



$${}^3J_{\text{H3-C2}} = 2.9 \text{ Hz}$$

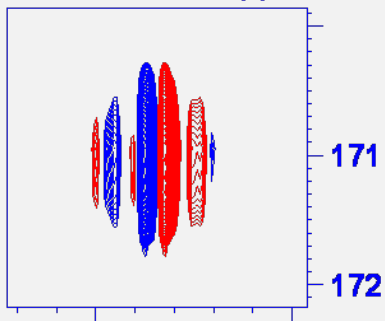
$${}^2J_{\text{H1-C2}} = 6.6 \text{ Hz}$$

$${}^2J_{\text{H4-C3}} = 4.5 \text{ Hz}$$

$${}^3J_{\text{H5-C3}} = 4.1 \text{ Hz}$$



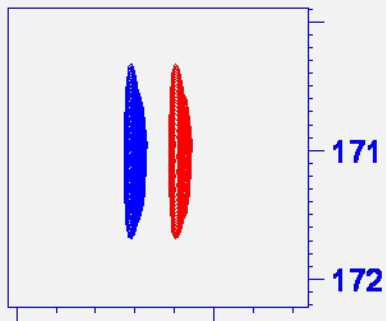
4.04 ppm



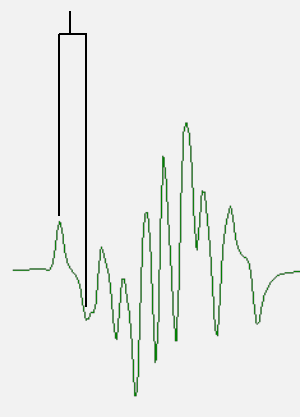
4.05



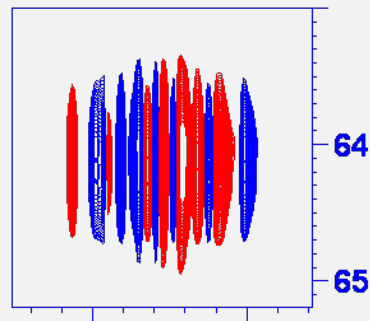
2.02 ppm



2.00



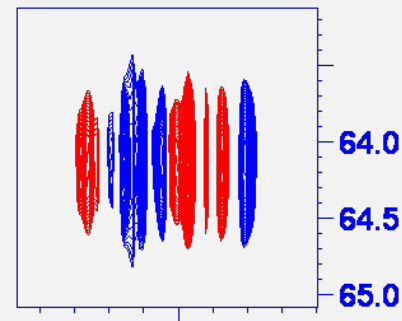
1.60 ppm



1.60 1.55



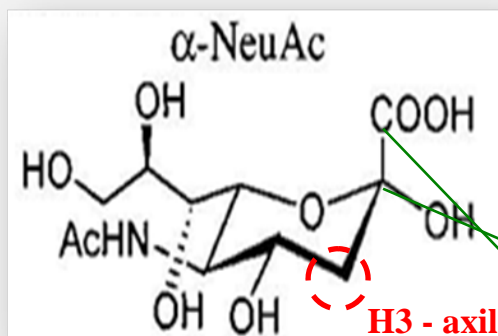
1.35 ppm



1.35

5.8 Sel_heteronuclear J-resolved

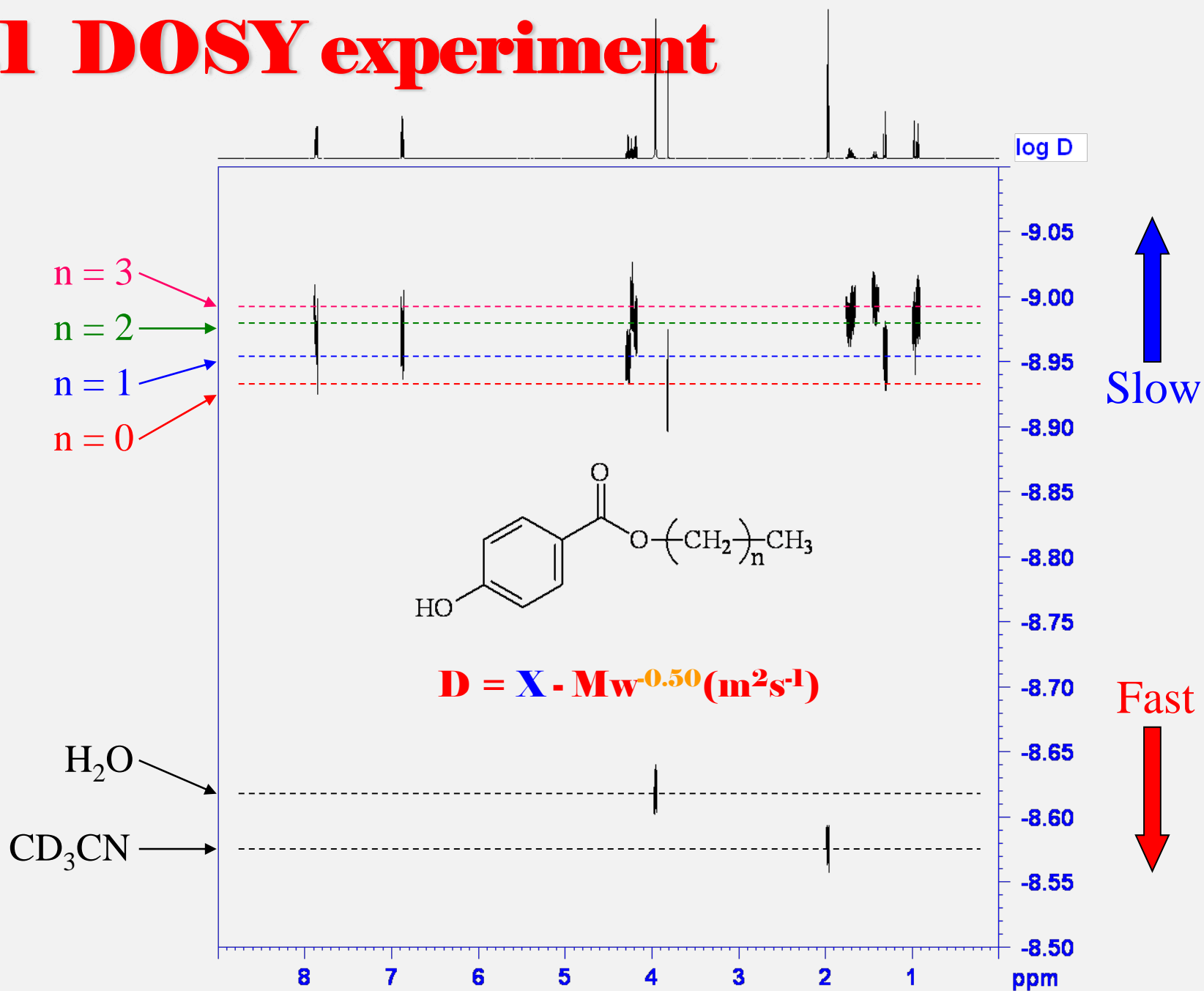
IGRC_2D_sel.HJres



6. DOSY (Diffusion-Ordered Spectroscopy) (擴散排序光譜)

- ☀ **DOSY experiment:** 區分混合物中物種數目
- ☀ 利用混合物中各成分之擴散係數不同(溶液中的移動速度)進行區分
- ☀ 擴散係數與分子之大小、形狀、環境溫度與溶劑之黏度有關
- ☀ 藉由擴散實驗區分在光譜中各訊號是否分屬於不同的分子

6.1 DOSY experiment



7. Other useful experiments

Selective - Selective Excitation

sel_COSY - NOESY

sel_TOCSY - TOCSY

sel_TOCSY - NOESY

sel_TOCSY - ROESY

sel_NOESY - TOCSY

sel_ROESY - TOCSY

sel_NOESY - NOESY

7.1

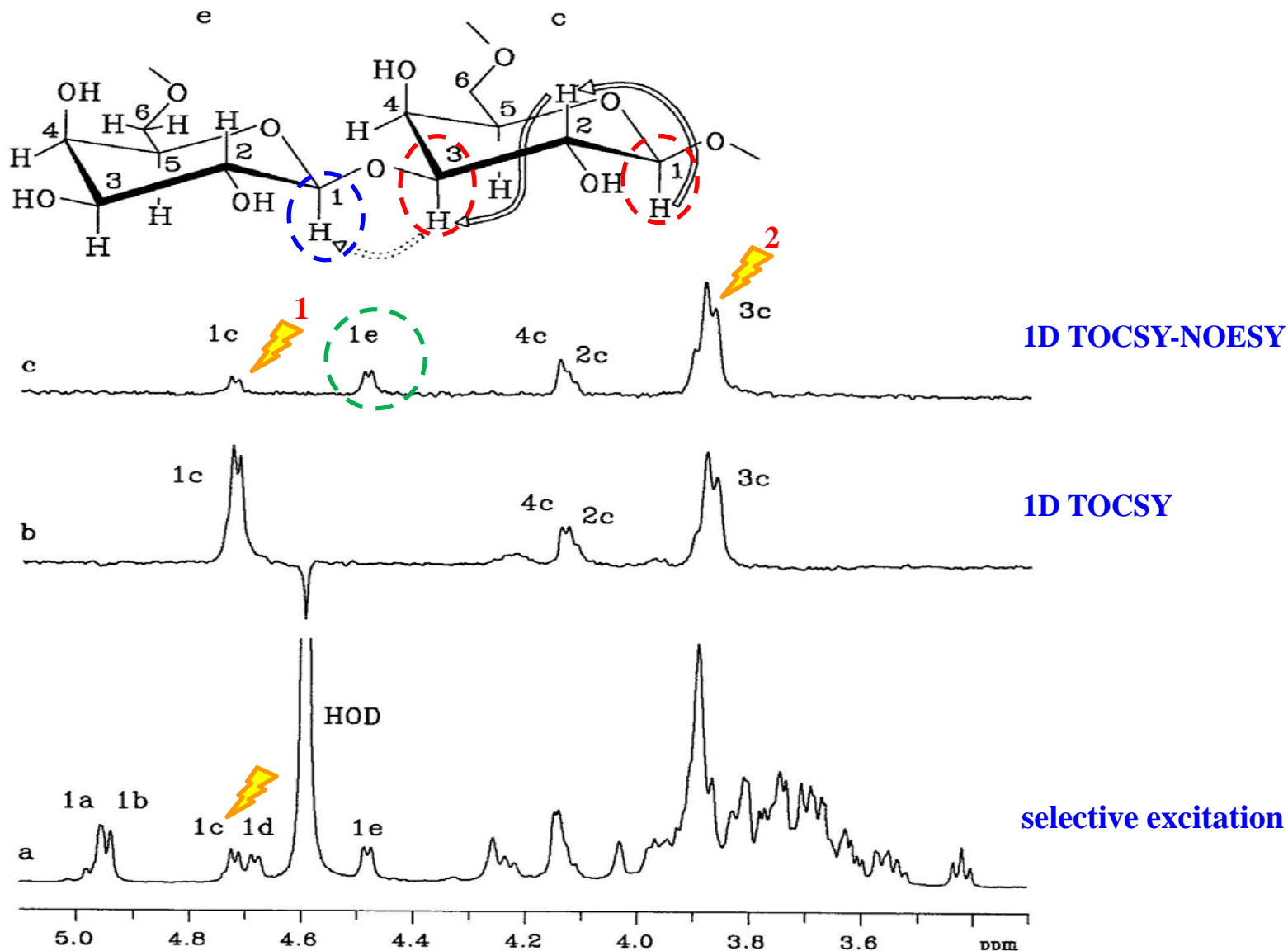


FIG. 5. Illustration of the 1D ge-TOCSY-NOESY technique on polysaccharide **3**. (a) ^1H spectrum of **3**, NT = 4; (b) 1D ge-TOCSY spectrum of **3**, NT = 8; H-1c was selectively excited by a 57 ms half-Gaussian pulse, the mixing time was 57.5 ms, and the δ delay was 28.8 ms; (c) 1D ge-TOCSY-NOESY spectrum (NT = 64) of **3** acquired using the pulse sequence of Fig. 1e. Selective pulses preceding the TOCSY and the NOESY periods were half-Gaussian pulses of 57 and 43.5 ms applied to H-1c and H-3c protons, respectively. Mixing times were 57.5 ms for the TOCSY and 250 ms for the NOESY transfer; the δ delay was 28.8 ms. A partial structure of **3** is given in the inset with TOCSY and NOESY pathways indicated by solid and dotted lines, respectively.

7.2

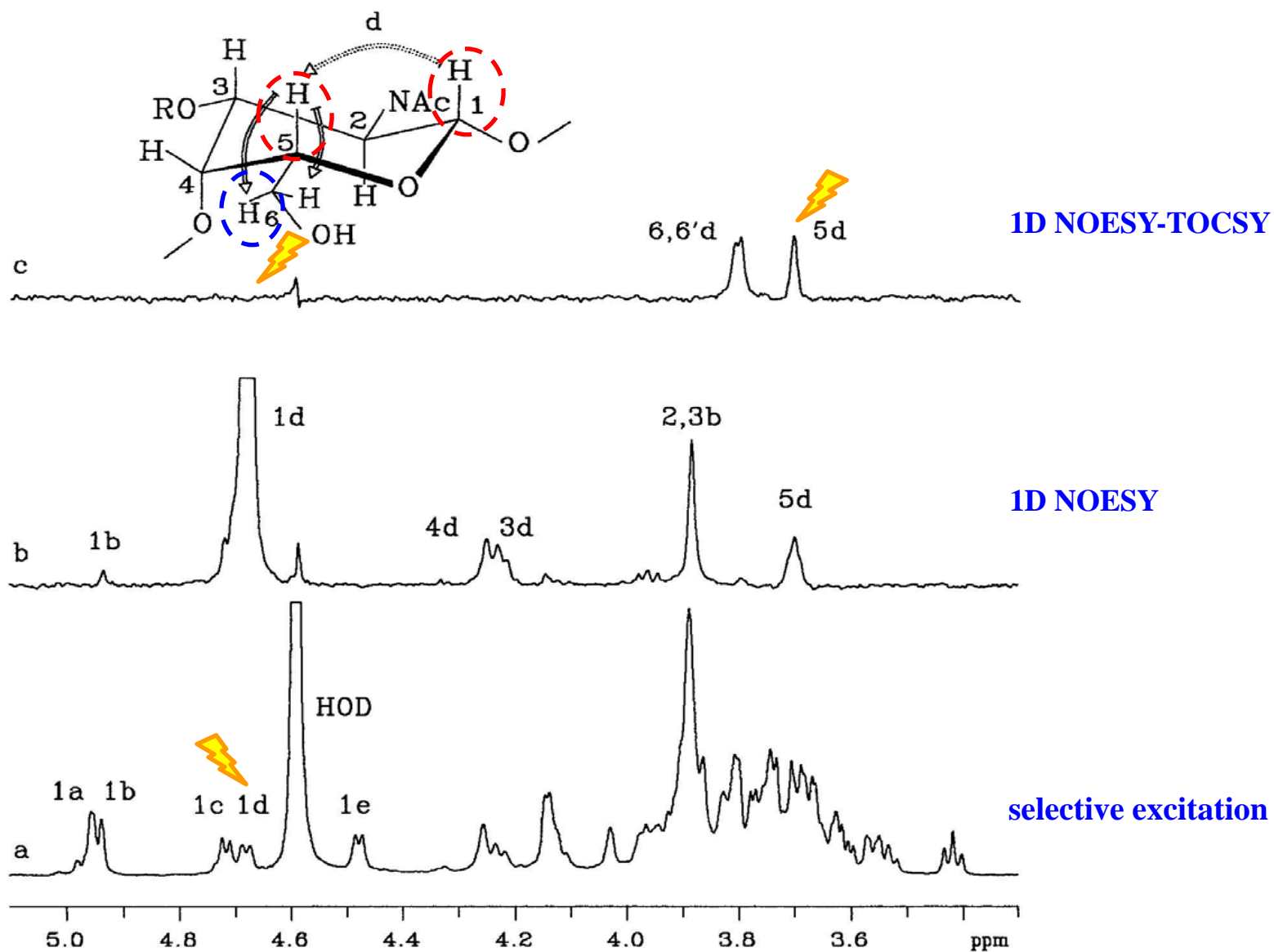


FIG. 6. Illustration of the 1D ge-NOESY-TOCSY technique on compound **3**. (a) ¹H spectrum of **3**, NT = 4; (b) 1D ge-NOESY spectrum of **3** (NT = 16); H-1d was selectively excited by a 57 ms half-Gaussian pulse; the mixing time was 250 ms. (c) 1D ge-NOESY-TOCSY spectrum (NT = 128) of **3** acquired using the pulse sequence of Fig. 1f. Selective pulses preceding the TOCSY and the NOESY period were half-Gaussian pulses of 57 and 43.5 ms applied to H-1d and H-5d protons, respectively. Mixing times were 250 ms for the NOESY and 48 ms for the TOCSY transfer; the δ delay was 24 ms. A partial structure of **3** is given in the inset with NOESY and TOCSY transfers indicated by dotted and solid lines, respectively.

Learning NMR

Reference :

☀ **NMR Guide**

☀ **The Hebrew University of Jerusalem**
(<http://chem.ch.huji.ac.il/nmr>)

☀ **Michigan State University**
(<http://www.cem.msu.edu/~reusch/VirtualText/Spectrpy/nmr/nmr1.htm>)

☀ **Department of Chemistry, Queen's University**
(<http://www.chem.queensu.ca/FACILITIES/NMR/nmr/webcourse/Flash/HX-coupling.swf>)

補1.

核種	自旋量子數	化學位移範圍	標準樣品
^1H	1/2	15 ~ -2	$\text{Si}(\text{CH}_3)_4$
^6Li	1	5 ~ -10	LiCl in H_2O
^7Li	3/2	5 ~ -10	LiCl in H_2O
^{13}C	1/2	250 ~ -20	$\text{Si}(\text{CH}_3)_4$
^{15}N	1/2	1200 ~ -500	CH_3NO_2
^{17}O	5/2	1800 ~ -100	H_2O
^{19}F	1/2	100 ~ -300	CFCl_3
^{23}Na	3/2	10 ~ -80	NaCl in H_2O
^{27}Al	5/2	300 ~ -300	$[\text{Al}(\text{H}_2\text{O})_6]^{3+}$
^{29}Si	1/2	100 ~ -400	$\text{Si}(\text{CH}_3)_4$
^{31}P	1/2	500 ~ -500	H_3PO_4

常用原子核的化學位移範圍值及其標準樣品

ACKNOWLEDGEMENTS

☀ 中研院基因體中心 張七鳳 博士

☀ 磊葳科技 吳英彥 博士

(資料的提供)

(技術的支援)

(知識的分享)